

Synthesis of Polarized Ethylenes Using Thioamides and Methyl  
Dithiocarboxylates and Their Application to Syntheses of Pyrazoles,  
Pyrimidines, Pyrazolo[3,4-*d*]pyrimidines, and 5-Aza[2.2.3]cyclazines  
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Polarized ethylenes having both electron-donating (an amino or a methylthio group) and electron-accepting (cyano, carbamoyl, methyl ester) groups on the adjacent two olefinic carbon atoms were prepared by the condensation of *S*-alkylthioamidinium salts or methyl dithiocarboxylates with the corresponding active methylene compounds in good yields. These polarized ethylenes were alternatively synthesized by the reaction of thioamides or methyl dithiocarboxylates with tetracyanoethylene oxide in good yields. Reactions of these polarized ethylenes with hydrazine or guanidine derivatives occurred smoothly to give the corresponding pyrazole and pyrimidine derivatives in good yields. The synthesis of 5-aza[2.2.3]cyclazine derivatives using polarized ethylenes is also described.

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Dithiocarboxylic acids and thioamides are most useful and successful in various synthetic applications of the thiocarbonyl derivatives [2-14]. In this paper, useful and convenient synthetic method of polarized ethylenes, which are important starting compounds for the synthesis of heterocyclic compounds, from thioamides and dithiocarboxylates is described, and furthermore the studies on synthesis of 5-aza[2.2.3]cyclazines are reported along with their some new reactions. Thioamides and methyl dithiocarboxylates using in this paper are shown in Figure 1.

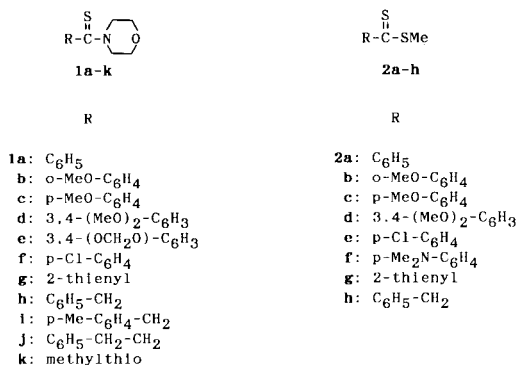


Fig. 1 Thioamides and dithiocarboxylates

### Synthesis of Polarized Ethylenes.

Ethylenes having electron-donating groups on an olefinic carbon atom and electron-accepting groups on another carbon atom of the olefin are one of most important and interesting compounds from both synthetic and theoretical points of view [15] (Figure 2). Among these compounds, for example, ketene dithioacetals [16-20], ethoxymethylene compounds [21-23], and aminomethylene compounds

[24-26] are widely used for the preparation of heterocyclic compounds.

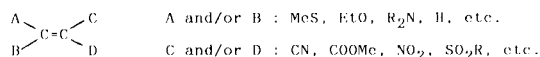
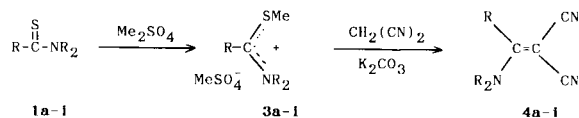


Fig. 2 Polarized ethylenes

Thioamides are easily activated with suitable alkylating reagents, giving the corresponding *S*-alkylthioamidinium salts. They are allowed to react efficiently with nucleophilic reagents such as Grignard reagents and amino compounds to give the corresponding displacement products of the alkylthio group [27-31]. However reactions of these *S*-alkylthioamidinium salts with active methylene compounds to give  $\beta$ -amino acrylonitrile derivatives are not almost known, and they are not sufficiently used in organic synthesis [27,28]. Therefore we attempted a synthesis of polarized ethylenes as electrophilic reagents.

Reactions of **3a-i**, readily prepared by treatment of thioamide derivatives **1a-i** with dimethyl sulfate without solvent, with active methylene compounds such as malononitrile in the presence of potassium carbonate in dimethyl sulfoxide gave the corresponding dicyanoethylene compounds **4a-i** in good yields (Scheme 1, Table 1).



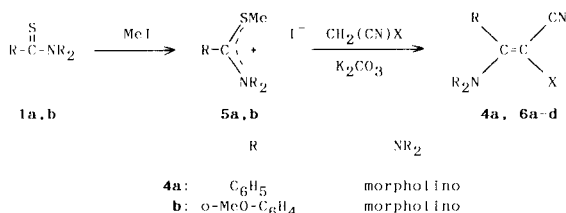
Scheme 1

The displacement reaction of methylthioamidinium iodides **5a,b**, which were prepared by treatment of thio-

Table 1. 3-Amino-2-cyanoacrylonitriles

	R	mp(°C)	Yield(%)
4a	C <sub>6</sub> H <sub>5</sub>	178	79
b	o-MeO-C <sub>6</sub> H <sub>4</sub>	107	87
c	p-MeO-C <sub>6</sub> H <sub>4</sub>	114	60
d	3,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	151	43
e	3,4-(OCH <sub>2</sub> O)-C <sub>6</sub> H <sub>3</sub>	187	57
f	p-Cl-C <sub>6</sub> H <sub>4</sub>	153	88
g	thienyl	145	83
h	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	113	59
i	p-Me-C <sub>6</sub> H <sub>4</sub> -CH <sub>2</sub>	120	58

amide derivatives **1a,b** with methyl iodide, with active methylene compounds (malononitrile, methyl cyanoacetate, cyanoacetamide) smoothly occurred in the presence of potassium carbonate in DMSO to give the corresponding polarized ethylenes **4a** and **6a-d**, though the yields were not so good (Scheme 2, Table 2).



Scheme 2

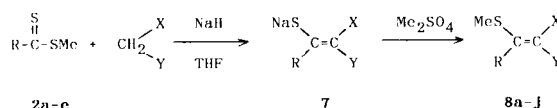
Table 2. Synthesis of Polarized Ethylenes

Entry	Product	R	X	Conditions <sup>a</sup> Time(hr), Solvent	mp(°C)	Yield(%) <sup>b</sup>
1	4a	Ph	CN	24 CH <sub>2</sub> Cl <sub>2</sub>	178	54
2	4a	Ph	CN	1 DMSO	178	46
3	6a	Ph	COOMe	24 CH <sub>2</sub> Cl <sub>2</sub>	116	43
4	b	Ph	CONH <sub>2</sub>	24 CH <sub>2</sub> Cl <sub>2</sub>	182	16
5	c	o-MeO-C <sub>6</sub> H <sub>4</sub>	COOMe	24 CH <sub>2</sub> Cl <sub>2</sub>	117	35
6	d	o-MeO-C <sub>6</sub> H <sub>4</sub>	CONH <sub>2</sub>	36 CH <sub>2</sub> Cl <sub>2</sub>	140	19

<sup>a</sup>All reactions were carried out at room temperature. <sup>b</sup>Yield after isolation by recrystallization.

Next, we explored a synthetic method of the new polarized ethylenes bearing methylthio groups as the electron-donating group. In general, this type of polarized ethylene derivatives are synthesized by the displacement reaction of suitable Grignard reagents with appropriately functionalized ketene dithioacetals, extensively used in the synthesis of heterocyclic compounds [16-20, 32-35].

Methyl dithiocarboxylates **2a-e**, which are readily prepared from the corresponding thioamides or Grignard reagents and carbon disulfide [36], are allowed to react with malononitrile in the presence of sodium hydride in tetrahydrofuran at reflux followed by the methylation with methyl iodide to give the corresponding 1-methylthio-2,2-dicyanoethylene derivatives **8a-e** in moderate yields (Scheme 3). Reactions of **2a-e** with other active methylene compounds (cyanoacetamide, methyl cyanoacetate, phenylacetone) also afforded the corresponding polarized ethylene compounds **8f-j** in 13-86% yields.

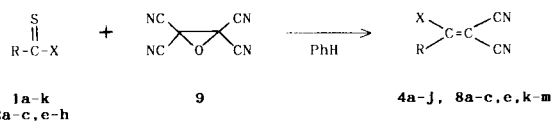


Scheme 3

Table 3. 2-Cyano-3-methylthioacrylonitriles

	R	X	Y	mp(°C)	Yield(%)
8a	C <sub>6</sub> H <sub>5</sub>	CN	CN	95	51
b	o-MeO-C <sub>6</sub> H <sub>4</sub>	CN	CN	75	87
c	p-MeO-C <sub>6</sub> H <sub>4</sub>	CN	CN	102	46
d	3,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	CN	CN	90	30
e	p-Cl-C <sub>6</sub> H <sub>4</sub>	CN	CN	155	56
f	C <sub>6</sub> H <sub>5</sub>	CN	CONH <sub>2</sub>	207	38
g	p-MeO-C <sub>6</sub> H <sub>4</sub>	CN	CONH <sub>2</sub>	130	55
h	p-Cl-C <sub>6</sub> H <sub>4</sub>	CN	CONH <sub>2</sub>	196	64
i	p-Me <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub>	CN	COOMe	120	13
j	p-Me <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub>	CN	C <sub>6</sub> H <sub>5</sub>	112	86

Moreover we found a novel and simple preparation of polarized ethylenes bearing push-pull substituents by the reaction of thioamides **1a-k** or methyl dithiocarboxylates **2a-c,e-h** with tetracyanoethylene oxide (**9**). Although it has been reported that the reaction of thiocarbonyl derivatives with tetracyanoethylene oxide give stable thiocarbonyl ylides, thiazoles, and dicyanomethylene compounds, the preparation of the polarized ethylenes using dicyanomethylenation with tetracyanoethylene oxide has been unknown hitherto [17-41]. Thioamides **1a-k** were allowed to react with tetracyanoethylene oxide at room temperature in benzene with stirring to give the corresponding dicyanoethylene compounds **4a-j** very smoothly. In addition, reactions of methyl dithiocarboxylates **2a-c,e-h** with tetracyanoethylene oxide also occurred under the similar condi-



Scheme 4

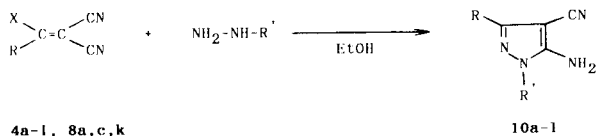
Table 4. Synthesis of 2-methylthio-1,1-dicyanoethylenes

Product	R	X	mp(°C)	% Yield
4a:	C <sub>6</sub> H <sub>5</sub>	morpholino	178	69
b:	o-MeO-C <sub>6</sub> H <sub>4</sub>	morpholino	95	91
c:	p-MeO-C <sub>6</sub> H <sub>4</sub>	morpholino	114	84
d:	3,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	morpholino	151	90
e:	3,4-(OCH <sub>2</sub> O)-C <sub>6</sub> H <sub>3</sub>	morpholino	187	82
f:	p-Cl-C <sub>6</sub> H <sub>4</sub>	morpholino	153	85
g:	2-thienyl	morpholino	145	88
h:	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	morpholino	113	77
i:	p-Me-C <sub>6</sub> H <sub>4</sub> -CH <sub>2</sub>	morpholino	120	75
j:	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> CH <sub>2</sub>	morpholino	111	79
k:	methylthio	morpholino	121	62
8a:	C <sub>6</sub> H <sub>5</sub>	methylthio	95	78
b:	o-MeO-C <sub>6</sub> H <sub>4</sub>	methylthio	75	67
c:	p-MeO-C <sub>6</sub> H <sub>4</sub>	methylthio	95	72
e:	p-Cl-C <sub>6</sub> H <sub>4</sub>	methylthio	155	82
k:	p-Me <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub>	methylthio	116	91
l:	2-thienyl	methylthio	102	78
m:	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	methylthio	o11	68

tions to yield the corresponding 2-methylthio-1,1-dicyanoethylene derivatives **8a-c,e,k-m** in good yields (Scheme 4). The results are listed in Table 4.

### Synthesis of Pyrazoles and Pyrimidines.

It is well known that the reaction of polarized ethylenes like ketene dithioacetals with bifunctionalized amines such as hydrazine or amidine derivatives gives the corresponding pyrazole or pyrimidine derivatives [42-47]. The heterocyclic compounds, thus obtained, are not only interesting from a viewpoint of biological activities but also important and useful as the starting materials for the synthesis to other heterocyclic compounds. Therefore, we attempted to prepare pyrazole and pyrimidine derivatives using **4** and **8**. Reactions of **4a-i** with hydrazine hydrate gave the corresponding 3-substituted 5-aminopyrazole-4-carbonitrile derivatives **10a-h** on heating at 100° for 5 hours in good yield. In contrast, reactions of morpholino compounds **4a-i** with phenylhydrazine did not occur under the similar conditions. However this problem can be readily solved by use of methylthioethylene compounds **8a,c,k** instead of **4a-i** since dicyanoethylenes bearing the methylthio group is generally more active towards nucleophiles than those of the morpholino group. Namely, reactions of **8a,c,k** with phenylhydrazine gave the corresponding 5-amino-1-arylpyrazoles-4-carbonitriles **10j-m** at reflux for 1 hour in ethanol, as illustrated in Table 5 (Scheme 5).

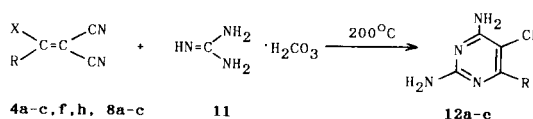


Scheme 5

Table 5. 5-Aminopyrazole-4-carbonitriles

Product	R	R'	mp(°C)	% Yield
<b>10a:</b>	C <sub>6</sub> H <sub>5</sub>	H	203	74
<b>b:</b>	o-MeO-C <sub>6</sub> H <sub>4</sub>	H	192	88
<b>c:</b>	p-MeO-C <sub>6</sub> H <sub>4</sub>	H	158	69
<b>d:</b>	3,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	H	175	51
<b>e:</b>	3,4-(OCH <sub>2</sub> O)-C <sub>6</sub> H <sub>3</sub>	H	215	65
<b>f:</b>	p-Cl-C <sub>6</sub> H <sub>4</sub>	H	221	82
<b>g:</b>	2-thienyl	H	255	91
<b>h:</b>	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	H	141	98
<b>i:</b>	p-Me-C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub>	H	162	98
<b>j:</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	160	46
<b>k:</b>	p-MeO-C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	190	43
<b>l:</b>	p-Cl-C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	189	64

Reactions of **4a-c,f,h** with guanidine carbonate (**11**) on heating at 200° for 2 hours gave the corresponding 2,4-diaminopyrimidine-5-carbonitrile derivatives **12a-e** in 52-90% yield (Scheme 6). The results are summarized in Table 6.



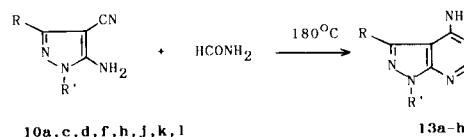
Scheme 6

Table 6. Synthesis of pyrazoles

Product	R	mp(°C)	% Yield
<b>12a:</b>	C <sub>6</sub> H <sub>5</sub>	300	85
<b>b:</b>	o-MeO-C <sub>6</sub> H <sub>4</sub>	218	70
<b>c:</b>	p-MeO-C <sub>6</sub> H <sub>4</sub>	245	83
<b>d:</b>	p-Cl-C <sub>6</sub> H <sub>4</sub>	281	90
<b>e:</b>	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	228	52

### Synthesis of Pyrazolo[3,4-d]pyrimidines.

Pyrazolo[3,4-d]pyrimidines are of considerable chemical and pharmacological importance, because of purine analogs [48]. In particular, hypoxanthine analogue, allopurinol, is a known inhibitor of xanthine oxidase in vivo [44]. Other related pyrazolo[3,4-d]pyrimidine, including the xanthine analogue (oxallopurinol) and the thiohypoxanthine analogue (thiopurinol), have similar inhibitory effect on xanthine oxidase. Many compounds of this type also reveal anti-tumor and anti-leukemic activity. Recently, several 3-substituted pyrazolo[3,4-d]pyrimidine derivatives have also shown pharmacological activity like allopurinol, which namely is the inhibitor of xanthine oxidase. Compounds **10** are the key intermediates for the preparation of pyrazolo[3,4-d]pyrimidine derivatives. It has been known the synthesis of 4-aminopyrazolo[3,4-d]pyrimidine by the cyclization of 5-aminopyrazole-4-carbonitrile with formamide [49,50]. So we tried the application of the above reaction to synthesis 3-substituted pyrazolo[3,4-d]pyrimidine derivatives. The reaction of **10a** with formamide also smoothly occurred at 180° for 2 hours to give the desired



Scheme 7

Table 7. 4-Aminopyrazolo[3,4-d]pyrimidines

	R	R'	mp(°C)	Yield(%)
<b>13a:</b>	C <sub>6</sub> H <sub>5</sub>	H	276	53
<b>b:</b>	p-MeO-C <sub>6</sub> H <sub>4</sub>	H	298	60
<b>c:</b>	3,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	H	290	42
<b>d:</b>	p-Cl-C <sub>6</sub> H <sub>4</sub>	H	357	50
<b>e:</b>	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	H	277	87
<b>f:</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	242	42
<b>g:</b>	p-MeO-C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	220	82
<b>h:</b>	p-Cl-C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	278	83

corresponding 4-amino-3-phenylpyrazolo[3,4-*d*]pyrimidine (**13a**) in good yields (Scheme 7). Other 3-substituted 4-aminopyrazolo[3,4-*d*]pyrimidine derivatives **3b-h** were also prepared from **10b,c,d,f,j-l** in a manner similar to that of the preparation of **13a** (Table 7).

It is well known that the reaction of *o*-aminobenzonitrile with carbon disulfide in the presence of an appropriate base like sodium hydroxide gives cyclized products, quinoxaline derivatives [57,64]. We tried the application of above reaction to synthesis of pyrazolo[3,4-*d*]pyrimidines. Compounds **10a-c,h** were allowed to react with carbon disulfide in the presence of potassium hydroxide as a base in dimethyl sulfoxide (DMSO), followed by treatment with methyl iodide to give the desired product, 4,6-bis(methylthio)pyrazolo[3,4-*d*]pyrimidines **14a-d** in good yields. Treatment of **14a-c** with 30% hydrogen peroxide in acetic acid afforded 4,6-hydroxy(or 4,6-dioxo)pyrazolo[3,4-*d*]pyrimidines **15a-c** in quantitative yields. The displacement of the methylthio group in **12a,b** with benzylamine smoothly occurs to yield the corresponding 4-benzylaminopyrazolo[3,4-*d*]pyrimidine derivatives **16a,b** in good yields (Scheme 8). In general, the 4-chloro group on the pyrimidine ring is more reactive than that of the 2-chloro group in the 2,4-dichloropyrimidine derivatives [42,51,52]. So it is suggested that the methylthio group at the 4-position in the pyrimidine is more active than that at the 2-position towards nucleophiles like amines [53].

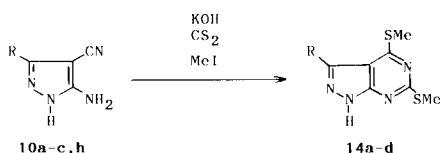
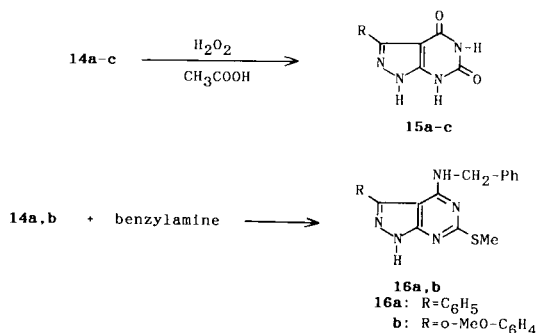


Table 8. 4,6-Bis(methylthio)pyrazolo[3,4-*d*]pyrimidines

	R	mp (°C)	Yield (%)
<b>14a</b>	C <sub>6</sub> H <sub>5</sub>	218	77
<b>b</b>	<i>o</i> -MeO-C <sub>6</sub> H <sub>4</sub>	200	64
<b>c</b>	<i>p</i> -MeO-C <sub>6</sub> H <sub>4</sub>	248	81
<b>d</b>	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub>	176	60

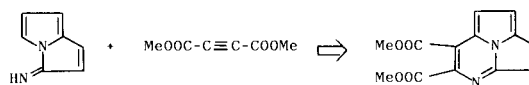


Scheme 8

### Synthesis of 5-Aza[2.2.3]cyclazines.

5-Aza[2.2.3]cyclazine is an interesting aromatic compound involving delocalized 10 $\pi$ -electrons from the theoretical point of view, similarly to other 1-, 2-, or 6-aza[2.2.3]cyclazines and [2.2.3]cyclazines [54-58]. Therefore synthesis of a variety of 5-azacyclazines by use of readily available starting materials is highly desirable for the investigation of aza[2.2.3]cyclazine chemistry. Boekelheide and Kertelj have first reported the synthesis of 5-aza[2.2.3]cyclazine derivatives by the [8 + 2] cycloaddition reaction of 7-methyl-2-phenylpyrrolo[1,2-*c*]pyrimidine with dimethyl acetylenedicarboxylate (DMAD) in the presence of palladium on charcoal, which contains a phenyl and a methyl group in the 6- and 2-positions, respectively [59]. However, access to desired 5-aza[2.2.3]cyclazine derivatives from pyrrolo[1,2-*c*]pyrimidines is difficult at present, because the preparation of suitably designed pyrrolo[1,2-*c*]pyrimidine derivatives is still limited.

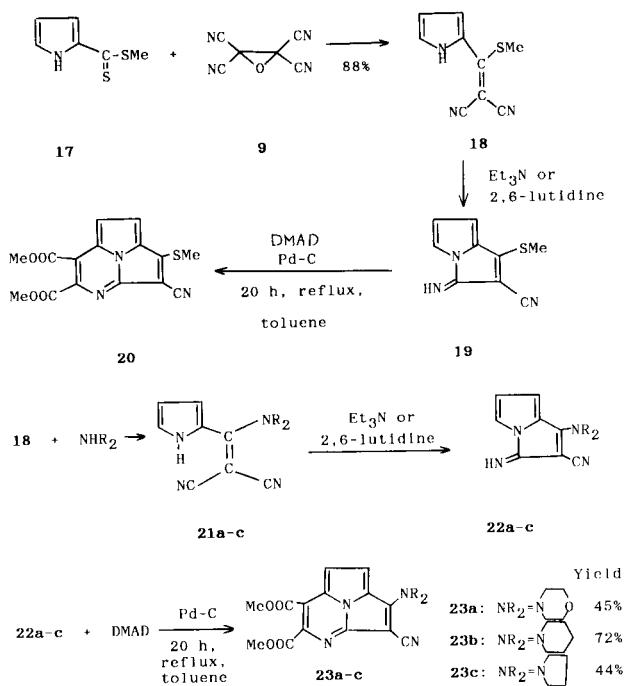
Jessep and Leaver reported a convenient and unique [8 + 2] cycloaddition reaction of 3-dimethylaminomethylene-3*H*-pyrrolizine bearing an exo-methylene group with DMAD to give dimethyl [2.2.3]cyclazine-5,6-dicarboxylate [60]. Therefore we attempted a novel synthesis of 5-aza[2.2.3]cyclazines by the [8 + 2] cycloaddition reaction of in situ generated cyclic exo-imino tetraene compounds, 3-imino-3*H*-pyrrolizine derivatives, with DMAD (Scheme 9).



Scheme 9

In an extension of the studies on polarized ethylenes [61], we applied this reaction to the preparation of 2-cyano-3-pyrrol-2-yl)-3-methylthioacrylonitrile which is an important precursor for the synthesis of 5-aza[2.2.3]cyclazines. Methyl pyrrole-2-dithiocarboxylate (**17**), readily prepared from the Grignard reagent of pyrrole and carbon disulfide, is allowed to react with tetracyanoethylene oxide (**9**) in benzene with stirring at room temperature to give a desired dicyanomethylene derivative **18** in 88% yield. Cyclization of **18** to 2-cyano-3-imino-1-methylthio-3*H*-pyrrolizine (**19**) occurred smoothly, when the mixture of **18** and triethylamine or 2,6-lutidine was heated at reflux, and then the reaction temperature was elevated ultimately up to *ca.* 150° by removal of the base [62]. Without isolation of **19**, treatment of the reaction mixture with DMAD in the presence of 5% palladium on charcoal in toluene at reflux for 20 hours gave the [8 + 2] cycloaddition product, dimethyl 4-cyano-3-methylthio-5-aza[2.2.3]cyclazine-6,7-dicarboxylate (**20**), in 23% yield. 3-Amino-5-aza[2.2.3]cyclazine derivatives **23a-c** were also obtained from the corre-

sponding 1-amino-3-imino-3*H*-pyrrolizine derivatives **22a-c** and DMAD. Compounds **21a-c** were readily prepared by the displacement of the methylthio group of **18** with the corresponding amines such as morpholine, piperidine, and pyrrolidine, respectively, followed by cyclization at heating in the presence of triethylamine or 2,6-lutidine in a manner similar to the preparation of **19** (Scheme 10). The vicinal coupling constant between C<sub>1</sub>-H and C<sub>2</sub>-H (7.62 and 7.86 pm,  $J_{1,2} = 4.5$  Hz) of **20** in the <sup>1</sup>H nmr spectrum is similar to the corresponding value of 2-methylthio-1-aza[2.2.3]cyclazine (7.21 and 7.53 ppm,  $J_{3,4} = 4.4$  Hz). The chemical shift of the methyl protons of the methylthio group of **20** is also shown in low field region similar to that of the corresponding methylthio group in 2-methylthio-1-aza[2.2.3]cyclazine [63]. The nmr data apparently indicate that 5-aza[2.2.3]cyclazine derivatives **20** and **23a-c** are typical aromatic compounds.



In conclusion, the utility of the present method for preparing polarized ethylenes **4,6,8,18** which were otherwise inaccessible was mostly displaced by the ready availability of starting materials and simple manipulation of the conversion. The activated ethylenes **4,6,8** are also very useful reagents for the preparation of a variety of heterocyclic compounds such as pyrazoles and pyrimidines. The present synthetic method of 5-aza[2.2.3]cyclazine by the [8 + 2] cycloaddition of heterocycles bearing exo-imino group with DMAD will provide a convenient and useful method for the synthesis of various other aza[2.2.3]cyclazine derivatives.

## EXPERIMENTAL

All melting points were determined in a capillary tube and are uncorrected. Infrared (ir) spectra were recorded in potassium bromide pellets on JASCO IRA-2 spectrometer and SHIMADZU IR-460 and ultraviolet (uv) absorption spectra were determined in 95% ethanol on a Hitachi EP-S2 spectrometer. Nuclear magnetic resonance (nmr) spectra were obtained on JNM-PS-100(100 MHz), JNM-FX-90Q(90 MHz), and JNM-PMX-60SI(60 MHz) spectrometers with tetramethylsilane as an internal standard. Mass (ms) spectra were recorded on a JMS-01SG and JMS-303D mass spectrometers.

### 2-Cyano-3-morpholino-3-phenylacrylonitrile (**4a**).

A mixture of 1.04 g (5 mmoles) of phenylthiomorpholide (**1a**) and 1.26 g (10 mmoles) of dimethyl sulfate was heated at 100° for 30 minutes. This imidium salts **3a** was used in the next step without purification. This crude product **3a** was mixed with 0.66 g (10 mmoles) of malononitrile, 20 ml of dimethyl sulfoxide, and 1.38 g (10 mmoles) of potassium carbonate. This mixture was stirred at room temperature for 2 hours and stirring was continued at 70° for 20 minutes. After cooling to room temperature, the reaction mixture was poured into 200 ml of water. The resulting precipitate was collected by filtration and recrystallized from methanol to give 0.95 g (4.0 mmoles) of **4a** as colorless needles, mp 178°, in 79% yield. Compound **4a** was also prepared from **5a** in manner similar to that described for the synthesis of **6a** (See Table 2); ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2190 (CN), 1595 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 236 (4.01), 314 (4.16); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  3.70 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>O), 7.59 (5H, s, phenyl-H).

*Anal.* Calcd. for C<sub>14</sub>H<sub>13</sub>N<sub>3</sub>O: C, 70.27; H, 5.48; N, 17.56. Found: C, 70.49; H, 5.52; N, 17.50.

### 2-Cyano-3-(2-methoxyphenyl)-3-morpholinoacrylonitrile (**4b**).

This compound (1.17 g, 4.4 mmoles) was synthesized in 87% yield from *o*-methoxyphenylthiomorpholide (**1b**) (1.19 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 107°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2170 (CN), 1595 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 217 (4.17), 311 (4.28); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  3.71 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>O), 3.89 (3H, s, OMe), 6.95-7.66 (4H, m, phenyl-H).

*Anal.* Calcd. for C<sub>15</sub>H<sub>17</sub>N<sub>3</sub>O<sub>2</sub>: C, 66.90; H, 5.61; N, 15.61. Found: C, 66.88; H, 5.52; N, 15.66.

### 2-Cyano-3-(4-methoxyphenyl)-3-morpholinoacrylonitrile (**4c**).

This compound (0.81 g, 3 mmoles) was synthesized in 60% yield from *o*-methoxyphenylthiomorpholide (**1c**) (1.19 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 114°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2190 (CN), 1600 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 223 (4.13), 298 (4.28); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  3.75 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>O), 3.87 (3H, s, OMe), 7.01 (2H, d,  $J = 8.4$  Hz, 2,6-H or 3,5-H), 7.39 (2H, d,  $J = 8.4$  Hz, 2,6-H or 3,5-H).

*Anal.* Calcd. for C<sub>15</sub>H<sub>17</sub>N<sub>3</sub>O<sub>2</sub>: C, 66.90; H, 5.61; N, 15.61. Found: C, 66.78; H, 5.61; N, 15.55.

### 2-Cyano-3-(3,4-dimethoxyphenyl)-3-morpholinoacrylonitrile (**4d**).

This compound (0.64 g, 2.2 mmoles) was synthesized in 43% yield from 3,4-dimethoxyphenylthiomorpholide (**1d**) (1.34 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 151°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2180 (CN), 1595 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 230 (4.16), 311 (4.29);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.74 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 3.91 (6H, s, OMe), 6.88 (1H, s, 2-H), 7.00 (2H, s, 5,6-H).

*Anal.* Calcd. for C<sub>16</sub>H<sub>17</sub>N<sub>3</sub>O<sub>3</sub>: C, 64.20; H, 5.72; N, 14.04. Found: C, 64.25; H, 5.78; N, 13.97.

#### 2-Cyano-3-(3,4-methylenedioxyphenyl)-3-morpholinoacrylonitrile (**4e**).

This compound (0.81 g, 2.9 mmoles) was synthesized in 57% yield from 3,4-methylenedioxyphenylthiomorpholide (**1e**) (1.26 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 187°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2180 (CN), 1600 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 233 (4.12), 311 (4.32);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.73 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 6.08 (2H, s, O-CH<sub>2</sub>-O), 6.84 (1H, s, 2-H), 6.94 (2H, s, 5,6-H).

*Anal.* Calcd. for C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>: C, 63.60; H, 4.63; N, 14.83. Found: C, 62.81; H, 4.38; N, 14.64.

#### 3-(4-Chlorophenyl)-2-cyano-3-morpholinoacrylonitrile (**4f**).

This compound (1.20 g, 4.4 mmoles) was synthesized in 88% yield from 4-chlorophenylthiomorpholide (**1f**) (1.21 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 153°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$ ; 2190 (CN), 1588 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 250 (4.12), 311 (4.14);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.74 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 7.34 (2H, d, J = 8.4 Hz, 2,6-H or 3,5-H), 7.47 (2H, d, J = 8.4 Hz, 2,6-H or 3,5-H).

*Anal.* Calcd. for C<sub>14</sub>H<sub>12</sub>ClN<sub>3</sub>O: C, 61.43; H, 4.42; Cl, 12.95; N, 15.35. Found: C, 61.27; H, 4.32; Cl, 13.07; N, 15.35.

#### 2-Cyano-3-morpholino-3-(2-thienyl)acrylonitrile (**4g**).

This compound (1.02 g, 4.2 mmoles) was synthesized in 83% yield from 2-thienylthiomorpholide (**1g**) (1.07 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give yellow needles, mp 145°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2180 (CN), 1525 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 244 (3.82), 312 (4.14);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.75 (8H, bs, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 7.10-7.35 (1H, m, 4-H), 7.58-7.78 (2H, m, 3,5-H).

*Anal.* Calcd. for C<sub>12</sub>H<sub>11</sub>N<sub>3</sub>OS: C, 58.76; H, 4.52; N, 17.13; S, 13.07. Found: C, 58.52; H, 4.44; N, 16.94; S, 12.96.

#### 3-Benzyl-2-cyano-3-morpholinoacrylonitrile (**4h**).

This compound (0.75 g, 3.0 mmoles) was synthesized in 59% yield from benzylthiomorpholide (**1h**) (1.11 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 113°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2180 (CN), 1555 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 297 (4.31);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.58-3.61 (8H, m, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 4.00 (2H, s, CH<sub>2</sub>), 7.40

(5H, s, phenyl-H).

*Anal.* Calcd. for C<sub>15</sub>H<sub>15</sub>N<sub>3</sub>O: C, 71.13; H, 5.97; N, 16.59. Found: C, 71.02; H, 5.91; N, 16.61.

#### 2-Cyano-3-(4-methylbenzyl)-3-morpholinoacrylonitrile (**4i**).

This compound (0.78 g, 2.9 mmoles) was synthesized in 58% yield from 4-methylbenzylthiomorpholide (**1i**) (1.18 g, 5 mmoles) and malononitrile (0.66 g, 10 mmoles) in a manner similar to that described for the preparation of **4a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 120°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2180 (CN), 1545 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 297 (4.01);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.32 (3H, s, Me), 3.56-3.61 (8H, m, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 3.94 (2H, s, CH<sub>2</sub>), 6.96-7.35 (4H, m, 2,3,5,6-H).

*Anal.* Calcd. for C<sub>16</sub>H<sub>17</sub>N<sub>3</sub>O: C, 71.89; H, 6.41; N, 15.72. Found: C, 71.96; H, 6.36; N, 15.66.

#### Methyl 2-Cyano-3-morpholino-3-phenylacrylate (**6a**).

A mixture of 2.07 g (10 mmoles) of phenylthiomorpholide (**1a**) and 2.84 g (20 mmoles) of methyl iodide was refluxed in methanol for 1 hour. This imidium salt **5a** was used in the next step without purification because of sufficient pure. The imidium salt (1.1 g, 5 mmoles) was mixed with 0.5 g (5 mmoles) of methyl cyanoacetate, 20 ml of dichloromethane, and 2.8 g (20 mmoles) of potassium carbonate. After stirring at room temperature for 24 hours, the mixture was poured into 50 ml of water, the organic layer was extracted by 30 ml of dichloromethane. After removal of solvent, this product was recrystallized from methanol to give 0.585 g (2.2 mmoles) of **5a** as colorless needles, mp 116°, in 43% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2190 (CN), 1690 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 242 (3.96), 329 (4.09);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.28-3.96 (8H, m, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 3.78 (3H, s, OMe), 7.18-7.61 (5H, m, phenyl-H).

*Anal.* Calcd. for C<sub>15</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>: C, 66.16; H, 5.92; N, 10.29. Found: C, 65.94; H, 5.95; N, 10.27.

#### 2-Cyano-3-morpholino-3-phenylacrylamide (**6b**).

This compound (0.205 g, 0.8 mmole) was synthesized in 16% yield from phenylthiomorpholide (**1a**) (1.04 g, 5 mmoles) and 2-cyanoacetamide (0.42 g, 5 mmoles) in a manner similar to that described for the preparation of **6a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 182°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3360, 3185 (NH<sub>2</sub>), 2190 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 242 (4.03), 339 (4.05);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.26-3.88 (8H, m, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 5.61 (2H, bs, NH<sub>2</sub>), 7.26-7.65 (5H, m, phenyl-H).

*Anal.* Calcd. for C<sub>14</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub>: C, 65.35; H, 5.88; N, 16.33. Found: C, 65.33; H, 5.86; N, 16.36.

#### Methyl 2-Cyano-3-morpholino-3-(2-methoxyphenyl)acrylate (**6c**).

This compound (0.531 g, 1.75 mmoles) was synthesized in 35% yield from 2-methoxyphenylthiomorpholide (**1d**) (1.19 g, 5 mmoles) and methyl cyanoacetate (0.5 g, 5 mmoles) in a manner similar to that described for the preparation of **6a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 117°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2200 (CN), 1705 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 232 (4.01, shoulder), 260 (3.55, shoulder), 323 (4.16);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.20-3.98 (8H, m, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 3.76 (3H, s, COOMe), 3.87 (3H, s, 2-OMe), 6.92-7.59 (4H, m, phenyl-H).

*Anal.* Calcd. for  $C_{16}H_{18}N_2O_4$ : C, 63.56; H, 6.00; N, 9.27. Found: C, 63.78; H, 5.93; N, 9.12.

#### 2-Cyano-3-(2-methoxyphenyl)-3-morpholinoacrylamide (**6d**).

This compound (0.272 g, 0.95 mmole) was synthesized in 19% yield from 2-methoxyphenylthiomorpholide (**1b**) (1.19 g, 5 mmoles) and 2-cyanoacetamide (0.42 g, 5 mmoles) in a manner similar to that described for the preparation of **6a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 140°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  3395, 3180 (NH<sub>2</sub>), 2190 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 232 (4.05, shoulder), 255 (3.72, shoulder), 324 (4.12); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  3.12-3.98 (8H, m, N-(CH<sub>2</sub>-CH<sub>2</sub>)<sub>2</sub>-O), 3.88 (3H, s, OMe), 5.64 (2H, bs, NH<sub>2</sub>), 6.84-7.58 (4H, m, phenyl-H).

*Anal.* Calcd. for  $C_{15}H_{17}N_3O_3$ : C, 62.71; H, 5.96; N, 14.62. Found: C, 62.72; H, 5.94; N, 14.72.

#### 2-Cyano-3-methylthio-3-phenylacrylonitrile (**8a**).

Malononitrile (0.99 g, 15 mmoles) was added to a suspension of sodium hydride (50%, 0.72 g, 15 mmoles) in 50 ml of absolute tetrahydrofuran (THF) and the mixture was stirred for 30 minutes at room temperature. Methyl phenyldithiocarboxylate (**2a**) (1.68 g, 10 mmoles) was added to the above mixture. The whole was stirred for 1 hour and then refluxed for 2 hours. After evaporation of the THF, the residue was dissolved in 100 ml of water. Dimethyl sulfate (1.89 g, 15 mmoles) was added dropwise to the solution and the mixture was stirred for 2 hours at room temperature. The precipitate was collected by filtration and recrystallized from methanol to give 1.02 g (51%) of colorless needles, mp 95°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  2220 (CN), 1597 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 278 (3.86, shoulder), 322 (4.14); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.22 (3H, s, SMe), 7.47 (5H, s, phenyl-H).

*Anal.* Calcd. for  $C_{11}H_8N_2S$ : C, 65.97; H, 4.03; N, 13.99; S, 16.01. Found: C, 65.94; H, 3.99; N, 14.00; S, 16.19.

#### 2-Cyano-3-(2-methoxyphenyl)-3-methylthioacrylonitrile (**8b**).

This compound (1.97 g, 8.7 mmoles) was synthesized in 87% yield from methyl 2-methoxyphenyldithiocarboxylate (**2b**) (1.98 g, 10 mmoles), malononitrile (0.99 g, 15 mmoles) and dimethyl sulfate (1.89 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 75°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  2200 (CN), 1594 (C=C); uv (ethanol):  $\lambda$  max nm (insufficient solubility) 253, 318;  $\lambda$  min nm 237, 266; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.15 (3H, s, SMe), 3.89 (3H, s, OMe), 6.98-7.16 (3H, m, 3,5,6-H), 7.40-7.60 (1H, m, 4-H).

*Anal.* Calcd. for  $C_{12}H_8N_2OS$ : C, 62.60; H, 4.38; N, 12.17; S, 13.90. Found: C, 62.15; H, 4.34; N, 12.23; S, 13.77.

#### 2-Cyano-3-(4-methoxyphenyl)-3-methylthioacrylonitrile (**8c**).

This compound (1.04 g, 4.6 mmoles) was synthesized in 46% yield from methyl 4-methoxyphenyldithiocarboxylate (**2c**) (1.98 g, 10 mmoles), malononitrile (0.99 g, 15 mmoles) and dimethyl sulfate (1.89 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 95°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  2220 (CN), 1600 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 254 (3.76), 280 (3.78, shoulder), 318 (4.16); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.14 (3H, s, SMe), 3.87 (3H, s, OMe), 6.94-7.41 (4H, m, 2,3,5,6-H).

*Anal.* Calcd. for  $C_{12}H_{10}N_2OS$ : C, 62.59; H, 4.38; N, 12.16; S, 13.92. Found: C, 62.49; H, 4.44; N, 12.38; S, 13.57.

#### 2-Cyano-3-(3,4-dimethoxyphenyl)-3-methylthioacrylonitrile (**8d**).

This compound (0.78 g, 3 mmoles) was synthesized in 30% yield from methyl 3,4-dimethoxyphenyldithiocarboxylate (**2d**) (2.28 g, 10 mmoles), malononitrile (0.99 g, 15 mmoles) and dimethyl sulfate (1.89 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 90°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  2210 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 260 (3.89), 315 (4.16); <sup>1</sup>H nmr (deuteriochloroform): 2.34 (3H, s, SMe), 3.93 (3H, s, OMe), 3.95 (3H, s, OMe), 6.90-7.15 (3H, m, 2,5,6-H).

*Anal.* Calcd. for  $C_{13}H_{12}N_2O_2S$ : C, 59.99; H, 4.65; N, 10.77; S, 12.30. Found: C, 59.81; H, 4.66; N, 10.95; S, 12.35.

#### 3-(4-Chlorophenyl)-2-cyano-3-methylthioacrylonitrile (**8e**).

This compound (1.32 g, 5.6 mmoles) was synthesized in 56% yield from methyl 4-chlorophenyldithiocarboxylate (**2e**) (2.03 g, 10 mmoles), malononitrile (0.99 g, 15 mmoles) and dimethyl sulfate (1.89 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 155°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  2210 (CN), 1582 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 294 (4.04, shoulder), 324 (4.14); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.28 (3H, s, SMe), 7.32 (2H, d, J = 8.79 Hz, 2,6-H or 3,5-H), 7.54 (2H, d, J = 8.79 Hz, 2,6-H or 3,5-H).

*Anal.* Calcd. for  $C_{11}H_7ClN_2S$ : C, 56.29; H, 3.01; N, 11.94; S, 13.66. Found: C, 55.99; H, 3.00; N, 11.75; S, 13.69.

#### 2-Cyano-3-methylthio-3-phenylacrylamide (**8f**).

This compound (0.73 g, 3.76 mmoles) was synthesized in 38% yield from methyl phenyldithiocarboxylate (**2a**) and cyanoacetamide (1.26 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 209°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  3360, 3160 (NH), 2200 (CN), 1620 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 238 (4.25), 314 (3.69); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  1.88 (3H, s, SMe), 5.81 (2H, bs, NH<sub>2</sub>), 7.15-7.27 (2H, m, phenyl-), 7.46-7.54 (3H, m, phenyl-H).

*Anal.* Calcd. for  $C_{11}H_{10}N_2OS$ : C, 60.53; H, 4.62; N, 12.84; S, 14.81. Found: C, 60.30; H, 4.63; N, 12.78; S, 14.81.

#### 2-Cyano-2-(4-methoxyphenyl)-3-methylthioacrylamide (**8g**).

This compound (1.36 g, 5.5 mmoles) was synthesized in 55% yield from methyl *p*-methoxyphenyldithiocarboxylate (**2c**) (1.99 g, 10 mmoles) and cyanoacetamide (1.26 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 130°; ir (potassium bromide):  $\nu$  max  $cm^{-1}$  3402, 3280, 3210 (NH), 2200 (CN), 1665 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 243 (3.64), 313 (4.23); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  1.88 (3H, s, SMe), 3.87 (3H, s, OMe), 5.77 (2H, bs, NH<sub>2</sub>), 6.96-7.54 (4H, m, phenyl-H).

*Anal.* Calcd. for  $C_{12}H_{12}O_2N_2S$ : C, 58.05; H, 4.87; N, 11.28; S, 12.91. Found: C, 58.01; H, 4.86; N, 11.29; S, 12.71.

#### 3-(4-Chlorophenyl)-2-cyano-3-methylthioacrylamide (**8h**).

This compound (2.25 g, 6.4 mmoles) was synthesized in 64% yield from methyl 4-chlorophenyldithiocarboxylate (**2e**) (2.02 g, 10 mmoles) and cyanoacetate (1.26 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless

needles, mp 196°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3380, 3200 ( $\text{NH}_2$ ), 2205 (CN), 1680 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 245 (4.06), 313 (4.23);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  1.91 (2H, s, SMe 2/3), 2.06 (1H, s, SMe 1/3), 7.09-7.55 (4H, m, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_9\text{ClON}_2\text{S}$ : C, 52.28; H, 3.59; Cl, 14.03; N, 11.08; S, 12.69. Found: C, 52.12; H, 3.60; Cl, 13.86; N, 11.00; S, 12.51.

Methyl 2-Cyano-3-(4-*N,N*-dimethylaminophenyl)-3-methylthioacrylate (**8i**).

This compound (0.36 g, 1.3 mmoles) was synthesized in 13% yield from methyl 4-*N,N*-dimethylaminophenyldithiocarboxylate (**2f**) (2.11 g, 10 mmoles), methyl cyanoacetate (1.49 g, 15 mmoles) and dimethyl sulfate (1.89 g, 15 mmoles) in a manner similar to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 120°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2250 (CN), 1680 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 264 (4.25), 318 (4.20), 390 (4.01);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.02 (3H, s, SMe), 3.01 (6H, s,  $\text{NMe}_2$ ), 3.84 (3H, s, OMe), 6.71 (2H, d,  $J = 8.8$  Hz, 2,6-H), 7.15 (2H, d,  $J = 8.8$  Hz, 3,5-H).

*Anal.* Calcd. for  $\text{C}_{14}\text{H}_{16}\text{N}_2\text{O}_2\text{S}$ : C, 60.85; H, 5.84; N, 10.14; S, 11.60. Found: C, 60.50; H, 5.89; N, 9.84; S, 11.40.

3-(4-*N,N*-Dimethylphenyl)-3-methylthio-2-phenylacrylonitrile (**8j**).

This compound (2.53 g, 8.6 mmoles) was synthesized in 86% yield from methyl 4-*N,N*-dimethylaminophenyldithiocarboxylate (**2f**) (2.11 g, 10 mmoles), phenylacetonitrile (1.76 g, 15 mmoles) and dimethyl sulfate (1.89 g, 15 mmoles) in a similar manner to that described for the preparation of **8a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 112°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2180 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 263 (4.21), 317 (4.11), 380 (4.21);  $^1\text{H}$  nmr (deuteriochloroform): 2.18 (3H, s, SMe), 3.03 (6H, s,  $\text{NMe}_2$ ), 6.53 (2H, d,  $J = 9.0$  Hz, 2,6-H), 7.01 (2H, d,  $J = 9.0$  Hz, 3,5-H), 7.18 (5H, s, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{18}\text{H}_{18}\text{N}_2\text{S}$ : C, 73.43; H, 6.16; N, 9.52; S, 10.89. Found: C, 73.50; H, 6.23; N, 9.35; S, 10.83.

General Procedure for the Reaction of Thiocarbonyl Compounds with Tetracyanoethylene Oxide.

A freshly prepared tetracyanoethylene oxide [64] (0.346 g, 2.4 mmoles) was added dropwise during 10 minutes to a solution of (2 mmoles) of the thiocarbonyl compounds while the mixture was stirred at 0°. After stirring for 3-5 hours at room temperature, the precipitate that appeared was collected by filtration and was recrystallized from the appropriate solvent such as methanol or benzene to give the corresponding acrylonitrile compounds. The filtrate (benzene solution) was chromatographed on a column of aluminium oxide using benzene as an eluent to give the corresponding acrylonitrile compounds. If the desired compounds did not appear in the reaction mixture, the product was obtained by chromatography on an aluminium oxide column followed by the crystallization from an appropriate solvent such as methanol. The results are listed in Table 4.

2-Cyano-3-morpholino-5-phenylbut-2-enylacrylonitrile (**4j**).

This compound was synthesized in 79% from **1j** and **9** in a manner similar to the above described method. An analytical sample was recrystallized from methanol to give colorless

needles, mp 111°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2190 (CN), 1560 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 300 (4.28);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.86-2.95 (4H, m,  $\text{CH}_2\text{-CH}_2$ ), 3.61 (8H, s,  $\text{N-(CH}_2\text{-CH}_2)_2\text{-O}$ ), 7.20-7.37 (5H, s, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{16}\text{H}_{17}\text{N}_3\text{O}$ : C, 71.88; H, 6.41; N, 15.72. Found: C, 72.00; H, 6.42; N, 15.61.

2-Cyano-3-methylthio-3-morpholinoacrylonitrile (**4l**).

This compound was synthesized in 62% yield from **1k** and **9**. An analytical sample was recrystallized from methanol to give colorless needles, mp 121°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2200 (CN), 1520 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 277 (3.85), 316 (4.22);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.63 (3H, s, SMe), 3.85 (8H, s,  $\text{N-(CH}_2\text{-CH}_2)_2\text{-O}$ ).

*Anal.* Calcd. for  $\text{C}_9\text{H}_{11}\text{N}_3\text{OS}$ : C, 51.65; H, 5.30; N, 20.08; S, 15.32. Found: C, 51.67; H, 5.25; N, 20.04; S, 15.57.

2-Cyano-3-(4-*N,N*-dimethylaminophenyl)-3-methylthioacrylonitrile (**8k**).

This compound was synthesized in 91% from **2f** and **9**. An analytical sample was recrystallized from methanol to give colorless needles, mp 116°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2225 (CN), 1606 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 258 (4.09), 323 (4.05), 430 (4.43);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.38 (3H, s, SMe), 3.09 (6H, s,  $\text{NMe}_2$ ), 6.73 (2H, d,  $J = 8.79$  Hz, 2,6-H, or 3,5-H), 7.45 (2H, d,  $J = 8.79$  Hz, 2,6-H or 3,5-H).

*Anal.* Calcd. for  $\text{C}_{13}\text{H}_{13}\text{N}_3\text{S}$ : C, 64.17; H, 5.39; N, 17.27; S, 13.18. Found: C, 64.07; H, 5.40; N, 17.18; S, 13.01.

2-Cyano-3-methylthio-3-(2-thienyl)acrylonitrile (**8l**).

This compound was synthesized in 78% from **2g** and **9**. An analytical sample was recrystallized from methanol to give colorless needles, mp 102°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2200 (CN), 1520 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 222 (3.95), 348 (4.18);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.60 (3H, s, SMe), 7.25 (1H, m, 4H), 7.68-7.83 (2H, m, 3,5-H).

*Anal.* Calcd. for  $\text{C}_9\text{H}_7\text{N}_2\text{S}_2$ : C, 52.17; H, 2.90; N, 13.52; S, 31.04. Found: C, 52.40; H, 2.93; N, 13.58; S, 31.09.

3-Benzyl-2-cyano-3-methylthioacrylonitrile (**8m**).

This compound was synthesized in 68% yield from **2h** and **9**. An analytical sample was recrystallized from methanol to give colorless oil; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2200 (CN), 1508 (C=C); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 313 (4.22);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.50 (3H, s, SMe), 4.13 (2H, s,  $\text{CH}_2$ ), 7.10-7.50 (5H, m, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{10}\text{N}_2\text{S}$ : C, 67.26; H, 4.70; N, 13.27; S, 14.96. Found: C, 67.08; H, 4.64; N, 13.10; S, 14.97.

5-Amino-3-phenyl-1*H*-pyrazole-4-carbonitrile (**10a**).

A mixture of 2.39 g (10 mmoles) of **4a** and 0.75 g (15 mmoles) of hydrazine hydrate was heated at 100° for 1 hour. This reaction product was recrystallized from methanol to give 1.36 g (7.4 mmoles) of **10a** as colorless needles, mp 203°, in 74% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3170 ( $\text{NH}_2$ , w), 2210 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 232 (4.25), 252 (4.14, shoulder);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  6.17 (2H, bs,  $\text{NH}_2$ ), 7.33-7.53 (3H, m, 3,4,5-H), 7.70-7.90 (2H, m, 2,6-H).

*Anal.* Calcd. for  $\text{C}_{10}\text{H}_8\text{N}_4$ : C, 65.21; H, 4.38; N, 30.42. Found: C, 65.49; H, 4.38; N, 30.47.

5-Amino-3-(2-methoxyphenyl)-1*H*-pyrazole-4-carbonitrile (**10b**).



This compound (1.89 g, 8.8 mmoles) was synthesized in 88% yield from **4b** (2.69 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a manner similar to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 192°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3200 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol): 253 (3.99), 293 (3.85);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.81 (3H, s, OMe), 5.84 (2H, bs,  $\text{NH}_2$ ), 6.91-7.55 (4H, m, 3,4,5,6-H).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_{10}\text{N}_4\text{O}$ : C, 61.67; H, 4.71; N, 26.16. Found: C, 61.79; H, 4.63; N, 26.05.

#### 5-Amino-3-(4-methoxyphenyl)-1H-pyrazole-4-carbonitrile (**10c**).

This compound (1.48 g, 6.9 mmoles) was synthesized in 69% yield from **3c** (2.69 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a manner similar to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 158°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3180 ( $\text{NH}_2$ , w), 2210 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 237 (4.01, shoulder), 264 (4.20);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.65 (3H, s, OMe), 6.12 (2H, s,  $\text{NH}_2$ ), 7.24 (2H, d, J = 9.6 Hz, 2,6-H or 3,5-H), 7.38 (2H, d, J = 9.6 Hz, 2,6-H or 3,5-H), 8.05 (1H, s, NH).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_{10}\text{N}_4\text{O}$ : C, 63.15; H, 5.30; N, 24.55. Found: C, 63.09; H, 5.26; N, 24.54.

#### 5-Amino-3-(3,4-dimethoxyphenyl)-1H-pyrazole-4-carbonitrile (**10d**).

This compound (1.25 g, 5.1 mmoles) was synthesized in 51% yield from **4d** (2.99 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a manner similar to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 175°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3230 ( $\text{NH}_2$ , w), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 268 (4.17), 292 (4.07, shoulder);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.83 (6H, s, OMe), 6.20 (2H, bs,  $\text{NH}_2$ ), 7.18 (1H, d, J = 7.8 Hz, 5-H or 6-H), 7.40 (1H, s, 2-H), 7.46 (1H, d, J = 7.8 Hz, 5-H or 6-H), 8.00 (1H, bs, NH).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{12}\text{N}_4\text{O}_2$ : C, 59.01; H, 4.95; N, 22.94. Found: C, 59.01; H, 4.89; N, 22.84.

#### 5-Amino-3-(3,4-methylenedioxyphenyl)-1H-pyrazole-4-carbonitrile (**10e**).

This compound (1.48 g, 6.5 mmoles) was synthesized in 65% yield from **4e** (2.83 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a manner similar to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 215°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3200 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 269 (4.00), 301 (3.95);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  6.07 (2H, s, O- $\text{CH}_2$ -O), 6.28 (2H, bs,  $\text{NH}_2$ ), 7.00 (1H, d, J = 9.0 Hz, 5-H or 6-H), 7.34 (1H, d, J = 9.0 Hz, 5-H or 6-H), 8.10 (1H, bs, NH).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_8\text{N}_4\text{O}_2$ : C, 57.89; H, 3.53; N, 24.55. Found: C, 57.70; H, 3.62; N, 24.17.

#### 5-Amino-3-(4-chlorophenyl)-1H-pyrazole-4-carbonitrile (**10f**).

This compound (1.79 g, 8.2 mmoles) was synthesized in 82% yield from **4f** (2.74 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a similar manner to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 221°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3180 ( $\text{NH}_2$ ), 2210 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 239 (4.26), 256 (4.23);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  6.54 (2H, bs,

$\text{NH}_2$ ), 7.56 (2H, d, J = 9.0 Hz, 2,6-H or 3,5-H), 7.86 (2H, d, J = 9.0 Hz, 2,6-H or 3,5-H).

*Anal.* Calcd. for  $\text{C}_{10}\text{H}_7\text{ClN}_4$ : C, 54.93; H, 3.23; N, 25.62; Cl, 16.21. Found: C, 54.94; H, 3.25; N, 25.35; Cl, 16.31.

#### 5-Amino-3-thien-2-yl-1H-pyrazole-4-carbonitrile (**10g**).

This compound (1.73 g, 9.1 mmoles) was synthesized in 91% yield from **4g** (2.45 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a manner similar to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give yellow needles, mp 255°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3400, 3320 (NH); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 225 (4.11), 262 (4.02), 278 (4.09);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide): 7.00-7.80 (5H, m, 3,4,5-H and NH).

*Anal.* Calcd. for  $\text{C}_8\text{H}_6\text{N}_4\text{S}$ : C, 50.51; H, 3.18; N, 29.45; S, 16.86. Found: C, 50.43; H, 3.18; N, 29.28; S, 16.89.

#### 5-Amino-3-benzyl-1H-pyrazole-4-carbonitrile (**10h**).

This compound (1.94 g, 9.8 mmoles) was synthesized in 98% yield from **4h** (2.53 g, 10 mmoles) and hydrazine hydrate (0.75 g, 10 mmoles) in a similar manner to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 141°; ir (potassium bromide):  $\delta$  max  $\text{cm}^{-1}$  3200 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 248 (3.73);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  6.40 (2H, bs,  $\text{NH}_2$ ), 7.15 (2H, s,  $\text{CH}_2$ ), 7.30 (5H, s, phenyl-H), 8.50 (1H, bs, NH).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_{10}\text{N}_4$ : C, 66.66; H, 5.09; N, 28.27. Found: C, 66.72; H, 5.17; N, 28.06.

#### 5-Amino-3-(4-methylbenzyl)-1H-pyrazole-4-carbonitrile (**10i**).

This compound (2.08 g, 9.8 mmoles) was synthesized in 98% yield from **4i** (2.67 g, 10 mmoles) and hydrazine hydrate (0.75 g, 15 mmoles) in a manner similar to that described for the preparation of **10a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 162°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3200 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 218 (4.24), 274 (2.77);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  2.22 (3H, s, Me), 3.74 (2H, s,  $\text{CH}_2$ ), 6.10 (2H, bs,  $\text{NH}_2$ ), 7.07 (4H, s, phenyl-H), 8.45 (1H, bs, NH).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{12}\text{N}_4$ : C, 67.91; H, 5.70; N, 26.40. Found: C, 67.79; H, 5.63; N, 26.00.

#### 5-Amino-1,3-diphenyl-1H-pyrazole-4-carbonitrile (**10j**).

A solution of 2.00 g (10 mmoles) of **8a** and 1.08 g (10 mmoles) of phenylhydrazine in 20 ml of ethanol was refluxed for 1 hour. After evaporation of the solvent, the residue was recrystallized from methanol to give 1.20 g (4.6 mmoles) of **10j** as colorless needles, mp 160°, in 46% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3440, 3340 ( $\text{NH}_2$ ), 2190 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 239 (4.44), 390 (2.85);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  6.80 (1H, bs,  $\text{NH}_2$ ), 7.55 (10H, s, phenyl-H), 7.80 (1H, bs,  $\text{NH}_2$ ).

*Anal.* Calcd. for  $\text{C}_{16}\text{H}_{12}\text{N}_4$ : C, 73.83; H, 4.65; N, 21.52. Found: C, 73.73; H, 4.59; N, 21.34.

#### 5-Amino-3-(4-methoxyphenyl)-1-phenylpyrazole-4-carbonitrile (**10k**).

This compound (1.25 g, 4.3 mmoles) was synthesized in 43% yield from **8c** (2.27 g, 10 mmoles) and phenylhydrazine (1.08 g, 10 mmoles) in a manner similar to that described for the preparation of **10i**. An analytical sample was recrystallized from methanol to give colorless needles, mp 190°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3430, 3325 ( $\text{NH}_2$ ), 2220 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ )

242 (4.43), 259 (4.40), 284 (4.22, shoulder);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.86 (3H, s, OMe), 4.57 (4H, bs,  $\text{NH}_2$ ), 6.96 (2H, d,  $J = 9.0$  Hz, 2,6-H), 7.54 (5H, m, phenyl-H), 7.93 (2H, d,  $J = 9.0$  Hz, 3,5-H).

*Anal.* Calcd. for  $\text{C}_{17}\text{H}_{14}\text{N}_4\text{O}$ : C, 70.33; H, 4.86; N, 19.30. Found: C, 70.32; H, 4.91; N, 19.27.

#### 5-Amino-3-(4-chlorophenyl)-1-phenylpyrazole-4-carbonitrile (**10f**).

This compound (1.89 g, 6.4 mmol) was synthesized in 64% yield from **8e** (2.35 g, 10 mmol) and phenylhydrazine (1.08 g, 10 mmol) in a manner similar to that described for the preparation of **10j**. An analytical sample was recrystallized from methanol to give colorless needles, mp 189°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3420, 3290 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 241 (4.44), 420 (2.75);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  6.95 (1H, bs,  $\text{NH}_2$ ), 7.35-7.80 (9H, m, phenyl-H), 8.18 (1H, bs,  $\text{NH}_2$ ).

*Anal.* Calcd. for  $\text{C}_{16}\text{H}_{11}\text{ClN}_4$ : C, 65.20; H, 3.76; N, 19.01. Found: C, 64.90; H, 3.72; N, 19.16.

#### 2,4-Diamino-6-phenylpyrimidine-4-carbonitrile (**12a**).

A mixture of 0.718 g (3 mmol) of **4a** and 0.541 g (3 mmol) of guanidine carbonate (**11**) was heated at 200° for 3 hours. After cooling, this product was washed two times with 10 ml of hot water, dried in air, and recrystallized from a mixture of methanol and benzene to give 0.539 g (2.6 mmol) of **12a** as colorless needles, mp 300°, in 85% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3450, 3380 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 249 (4.47), 311 (3.85);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  6.28 (2H, bs,  $\text{NH}_2$ ), 6.50 (2H, bs,  $\text{NH}_2$ ), 7.22-7.33 (5H, m, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_9\text{N}_5$ : C, 62.55; H, 4.29; N, 33.16. Found: C, 62.00; H, 4.20; N, 33.29.

#### 2,4-Diamino-6-(2-methoxyphenyl)pyrimidine-5-carbonitrile (**12b**).

This compound (0.507 g, 2.1 mmol) was synthesized in 70% yield from **4b** (0.808 g, 3 mmol) and guanidine carbonate (**11**) (0.541 g, 3 mmol) in a manner similar to that described for the preparation of **12a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 218°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3500, 3350 ( $\text{NH}_2$ ), 2210 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 251 (4.32), 300 (4.00).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{11}\text{N}_5\text{O}$ : C, 55.59; H, 5.05; N, 27.02. Found: C, 55.28; H, 4.75; N, 26.86.

#### 2,4-Diamino-6-(4-methoxyphenyl)pyrimidine-5-carbonitrile (**12c**).

This compound (0.601 g, 2.5 mmol) was synthesized in 83% yield from **4c** (0.808 g, 3 mmol) and guanidine carbonate (**11**) (0.541 g, 3 mmol) in a manner similar to that described for the preparation of **12a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 245°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3420, 3380 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 259 (4.43), 286 (4.22), 320 (3.89, shoulder);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.85 (3H, s, OMe), 6.98 (4H, s,  $\text{NH}_2$ ), 7.04 (2H, d,  $J = 9.0$  Hz, 3,5-H or 2,6-H), 7.82 (2H, d,  $J = 9.0$  Hz, 3,5-H or 2,6-H).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{11}\text{N}_5\text{O}$ : C, 59.74; H, 4.60; N, 29.03. Found: C, 59.34; H, 4.58; N, 28.68.

#### 2,4-Diamino-6-(4-chlorophenyl)pyrimidine-5-carbonitrile (**12d**).

This compound (0.666 g, 2.7 mmol) was synthesized in 90% yield from **4f** (0.821 g, 3 mmol) and guanidine carbonate (0.541

g, 3 mmol) in a similar manner to that described for the preparation of **12a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 281°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3430, 3375 ( $\text{NH}_2$ ), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 253 (4.54), 314 (3.86);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.55 (4H, bs,  $\text{NH}_2$ ), 6.88-7.27 (4H, m, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_8\text{ClN}_5$ : C, 53.80; H, 3.28; N, 28.52. Found: C, 53.56; H, 3.18; N, 28.55.

#### 2,4-Diamino-6-benzylpyrimidine-5-carbonitrile (**12e**).

This compound (0.428 g, 1.6 mmol) was synthesized in 52% yield from **4h** (0.760 g, 3 mmol) and guanidine carbonate (0.541 g, 3 mmol) in a manner similar to that described for the preparation of **12a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 228°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3365, 3325 ( $\text{NH}_2$ ), 2205 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 242 (4.23), 250 (4.23, shoulder), 296 (3.88), 320 (3.37, shoulder);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.95 (2H, s,  $\text{CH}_2$ ), 5.25 (4H, bs,  $\text{NH}_2$ ), 7.13-7.31 (5H, m, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{11}\text{N}_5$ : C, 63.99; H, 4.92; N, 31.09. Found: C, 63.77; H, 4.88; N, 31.07.

#### 4-Amino-3-phenylpyrazolo[3,4-*d*]pyrimidine (**13a**).

A mixture of 2.0 g (10.9 mmol) of **10a** and 2 ml of formamide was heated at 200° for 2 hours. After cooling, the solid was washed with 20 ml of water and recrystallized from a mixture of methanol and benzene to give 1.21 g (5.8 mmol) of **13a**, colorless needles, mp 276°, in 53% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3160 ( $\text{NH}_2$ ); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 254 (4.18), 276 (4.12, shoulder);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  6.68 (2H, bs,  $\text{NH}_2$ ), 7.60 (5H, s, phenyl-H), 8.27 (1H, s, NH).

*Anal.* Calcd. for  $\text{C}_{11}\text{H}_9\text{N}_5$ : C, 62.55; H, 4.30; N, 33.16. Found: C, 62.22; H, 4.23; N, 32.83.

#### 4-Amino-3-(4-methoxyphenyl)pyrazolo[3,4-*d*]pyrimidine (**13b**).

This compound (1.35 g, 5.6 mmol) was synthesized in 60% yield from **10c** (2.0 g, 9.3 mmol) and formamide (2 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 298°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3420 ( $\text{NH}_2$ ); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 262 (4.20), 280 (4.11, shoulder);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  3.88 (3H, s, OMe), 6.70 (2H, bs,  $\text{NH}_2$ ), 7.16 (2H, d,  $J = 7.2$  Hz, 2,6-H), 7.55 (2H, d,  $J = 7.2$  Hz, 3,5-H), 8.26 (1H, s, NH).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{11}\text{N}_5\text{O}$ : C, 59.74; H, 4.60; N, 29.03. Found: C, 59.66; H, 4.62; N, 28.93.

#### 4-Amino-3-(3,4-dimethoxyphenyl)pyrazolo[3,4-*d*]pyrimidine (**13c**).

This compound (0.93 g, 3.4 mmol) was synthesized in 42% yield from **10d** (2.0 g, 8.2 mmol) and formamide (2 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 290°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3400 ( $\text{NH}_2$ ); uv (ethanol):  $\lambda$  max nm (insufficient solubility) 266, 285;  $\lambda$  min nm 250.

*Anal.* Calcd. for  $\text{C}_{13}\text{H}_{13}\text{N}_5\text{O}_2$ : C, 57.56; H, 4.83; N, 25.82. Found: C, 57.56; H, 4.86; N, 25.79.

#### 4-Amino-3-(4-chlorophenyl)pyrazolo[3,4-*d*]pyrimidine (**13d**).

This compound (1.11 g, 4.5 mmol) was synthesized in 50%

yield from **10f** (2.0 g, 9.1 mmoles) and formamide (2 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from a mixture of methanol and benzene to give colorless needles, mp 357°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3060 (NH<sub>2</sub>); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 257 (4.29), 280 (4.21, shoulder).

*Anal.* Calcd. for C<sub>11</sub>H<sub>8</sub>ClN<sub>5</sub>: C, 53.78; H, 3.28; Cl, 14.43; N, 28.51. Found: C, 54.01; H, 3.33; Cl, 14.31; N, 28.32.

#### 4-Amino-3-benzylpyrazolo[3,4-*d*]pyrimidine (**13e**).

This compound (1.04 g, 4.6 mmoles) was synthesized in 87% yield from **10h** (1.0 g, 5.3 mmoles) and formamide (1 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 277°; ir (potassium carbonate):  $\nu$  max  $\text{cm}^{-1}$  3455, 3305, 3080, 3005 (NH); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 260 (3.95), 277 (3.93); <sup>1</sup>H nmr (deuteriochloroform + trifluoroacetic acid 5:1):  $\delta$  4.47 (2H, s, CH<sub>2</sub>-Ph), 7.22-7.48 (5H, m, phenyl-H), 8.58 (1H, s, 6-H).

*Anal.* Calcd. for C<sub>12</sub>H<sub>11</sub>N<sub>5</sub>: C, 63.99; H, 4.92; N, 31.09. Found: C, 63.91; H, 5.02; N, 30.91.

#### 4-Amino-1,3-diphenylpyrazolo[3,4-*d*]pyrimidine (**13f**).

This compound (0.46 g, 1.60 mmoles) was synthesized in 42% yield from **10j** (1.0 g, 3.85 mmoles) and formamide (1 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 242°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3230 (NH); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 258 (4.81); <sup>1</sup>H nmr (deuterio-dimethyl sulfoxide):  $\delta$  6.89-7.57 (10H, m, phenyl-H), 8.63 (1H, s, 6-H).

*Anal.* Calcd. for C<sub>17</sub>H<sub>13</sub>N<sub>5</sub>: C, 71.07; H, 4.56; N, 24.37. Found: C, 69.98; H, 4.41; N, 24.67.

#### 4-Amino-1-phenyl-3-(4-methoxyphenyl)pyrazolo[3,4-*d*]pyrimidine (**13g**).

This compound (0.89 g, 2.82 mmoles) was synthesized in 82% yield from **10k** (1.0 g, 3.45 mmoles) and formamide (1 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 220°; ir (potassium bromide):  $\lambda$  max  $\text{cm}^{-1}$  3450, 3060 (NH); uv (ethanol):  $\nu$  max nm (log  $\epsilon$ ) 242 (4.51), 299 (4.18); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  3.88 (3H, s, OMe), 5.52 (2H, bs, NH<sub>2</sub>), 7.18 (2H, d, J = 8.0 Hz, phenyl-H), 7.26-7.65 (5H, m, phenyl-H), 8.21 (2H, d, J = 8.0 Hz), 8.45 (1H, s, 6-H).

*Anal.* Calcd. for C<sub>18</sub>H<sub>15</sub>ON<sub>5</sub>: C, 68.13; H, 4.76; N, 22.07. Found: C, 68.13; H, 4.78; N, 22.23.

#### 4-Amino-3-(4-chlorophenyl)-1-phenylpyrazolo[3,4-*d*]pyrimidine (**13h**).

This compound (0.90 g, 2.96 mmoles) was synthesized in 83% yield from **10l** (1.0 g, 3.45 mmoles) and formamide (1 ml) in a manner similar to that described for the preparation of **13a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 278°; ir (potassium bromide):  $\lambda$  max  $\text{cm}^{-1}$  3480, 3060 (NH); uv (ethanol):  $\nu$  max nm (log  $\epsilon$ ) 242 (4.40), 302 (4.18); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  5.50 (2H, bs, NH), 7.26-7.79 (7H, m, phenyl-H), 8.19 (2H, d, J = 8.0 Hz, phenyl-H), 8.49 (1H, s, 6-H).

*Anal.* Calcd. for C<sub>17</sub>H<sub>12</sub>ClN<sub>5</sub>: C, 63.46; H, 3.76; N, 21.77. Found: C, 63.24; H, 3.73; N, 22.02.

#### 4,6-Bis(methylthio)-3-phenylpyrazolo[3,4-*d*]pyrimidine (**14a**).

To a solution of 0.92 g (5 mmoles) of **10a** and 20% solution of potassium hydroxide (KOH: 1.12 g, H<sub>2</sub>O: 4.5 ml) in 10 ml of dimethyl sulfoxide, stirred at 0°, 0.76 g (10 mmoles) of carbon disulfide was added in several portions during 30 minutes. After another 1 hour at room temperature, 2.13 g (15 mmoles) of methyl iodide was slowly added to the stirring solution over a period of 30 minutes and stirring was continued for 1 hour at room temperature. The reaction mixture was poured into 10 ml of ice-water. The precipitate that appeared was collected by filtration and washed several times with water. This crude product was recrystallized from methanol to give 1.11 g (3.9 mmoles) of colorless needles, mp 218°, in 77% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3185 (NH), 1580, 1531, 1229; uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 257 (4.20), 300 (3.78, shoulder); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.60 (3H, s, SMe), 2.66 (3H, s, SMe), 7.45-7.54 (3H, m, 3,4,5-H), 7.69-7.80 (2H, m, 2,6-H).

*Anal.* Calcd. for C<sub>13</sub>H<sub>12</sub>N<sub>4</sub>S<sub>2</sub>: C, 54.14; H, 4.19; N, 19.43; S, 22.24. Found: C, 54.05; H, 4.23; N, 19.56; S, 22.35.

#### 3-(2-Methoxyphenyl)-4,6-bis(methylthio)pyrazolo[3,4-*d*]pyrimidine (**14b**).

This compound (1.02 g, 3.2 mmoles) was synthesized in 64% yield from **10b** (1.07 g, 5 mmoles), carbon disulfide (0.76 g, 10 mmoles) and methyl iodide (2.13 g, 15 mmoles) in a manner similar to that described for the preparation of **14a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 200°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3155, 3000 (NH), 1585, 1538, 1259; uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 220 (4.30, shoulder), 254 (4.43), 300 (4.18); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.54 (3H, s, SMe), 2.70 (3H, s, SMe), 3.79 (3H, s, OMe), 6.96-7.57 (4H, m, phenyl-H).

*Anal.* Calcd. for C<sub>14</sub>H<sub>13</sub>N<sub>4</sub>OS<sub>2</sub>: C, 52.98; H, 4.13; N, 17.65; S, 20.20. Found: C, 52.63; N, 4.39; N, 17.61; S, 20.08.

#### 3-(4-Methoxyphenyl)-4,6-bis(methylthio)pyrazolo[3,4-*d*]pyrimidine (**14c**).

This compound (1.29 g, 4.1 mmoles) was synthesized in 81% yield from **10c** (1.07 g, 5 mmoles), carbon disulfide (0.76 g, 10 mmoles) and methyl iodide (2.13 g, 15 mmoles) in a manner similar to that described for the preparation of **14a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 248°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3190, 3035 (NH), 1590, 1538, 1244; uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 236 (4.33), 260 (4.56), 305 (4.20); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.60 (3H, s, SMe), 2.66 (3H, s, SMe), 3.88 (3H, s, OMe), 7.01 (2H, d, J = 8.79 Hz, 2,6-H or 3,5-H), 7.69 (2H, d, J = 8.79 Hz, 2,6-H or 3,5-H).

*Anal.* Calcd. for C<sub>14</sub>H<sub>13</sub>N<sub>4</sub>S<sub>2</sub>: C, 52.83; H, 4.43; N, 17.60; S, 20.11. Found: C, 52.73; H, 4.48; N, 17.63; S, 19.90.

#### 3-Benzyl-4,6-bis(methylthio)pyrazolo[3,4-*d*]pyrimidine (**14d**).

This compound (0.91 g, 3 mmoles) was synthesized in 60% yield from **10h** (0.99 g, 5 mmoles), carbon disulfide (0.76 g, 10 mmoles) and methyl iodide (2.13 g, 15 mmoles) in a manner similar to that described for the preparation of **14a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 176°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3185 (NH), 1762, 1634, 1597, 1527; uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 218 (4.26, shoulder), 256 (4.55, shoulder), 262 (4.57), 305 (3.93); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.60 (3H, s, SMe), 2.63 (3H, s, SMe), 4.37 (2H, s, -CH<sub>2</sub>-), 7.25 (5H, s, phenyl-H), 8.55 (1H, s, NH).

*Anal.* Calcd. for C<sub>14</sub>H<sub>14</sub>N<sub>4</sub>S<sub>2</sub>: C, 55.60; H, 4.67; N, 18.53; S, 21.20. Found: C, 55.69; H, 4.68; N, 18.55; S, 21.23.

4,6-Dihydroxy-3-phenylpyrazolo[3,4-*d*]pyrimidine (**15a**).

A solution of 0.865 g (3 mmoles) of **14a**, 10 ml of 30% hydrogen peroxide in 30 ml of acetic acid was heated at 70° for 5 hours. After evaporation of the solvent, the residue was washed with 10 ml of water, the product was dried in air and recrystallized from methanol to give 0.602 g (2.6 mmoles) of **15a** as colorless needles, mp > 384° [lit, [44], mp > 370°], in 88% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3495, 3195, 3075 (OH or NH), 1750, 1661 (CO);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  7.44-7.56 (3H, m, phenyl-H), 8.05-8.19 (2H, m, phenyl-H), 10.57 (1H, bs, NH or OH), 11.32 (1H, bs, NH or OH); ms:  $m/z$  228 ( $\text{M}^+$ , 77), 195 (11), 42 (100).

4,6-Dihydroxy-3-(2-methoxyphenyl)pyrazolo[3,4-*d*]pyrimidine (**15b**).

This compound (0.689 g, 2.7 mmoles) was synthesized in 89% yield from **14b** (0.954 g, 3 mmoles) and 30% hydrogen peroxide (10 ml) in a manner similar to that described for the preparation of **15a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 323°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3410, 3235, 3080 (OH or NH), 1718, 1670, 1605 (CO);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  3.82 (3H, s, SMe), 7.03 (1H, dt,  $J = 1.1, 7.2$  Hz, 5'-H), 7.14 (1H, d,  $J = 7.0$  Hz, 3'-H), 7.45 (1H, m, 4'-H), 7.84 (1H, d,  $J = 7.5$  Hz, 6'-H), 10.43 (1H, bs, NH or OH), 11.20 (1H, bs, NH or OH), 13.06 (1H, bs, NH or OH); ms:  $m/z$  258 ( $\text{M}^+$ , 77), 229 (11), 215 (17), 197 (30), 42 (100).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{10}\text{N}_4\text{O}_3$ : C, 55.81; H, 3.50; N, 21.70. Found: C, 55.67; H, 3.49; N, 21.45.

4,6-Dihydroxy-3-(4-methoxyphenyl)pyrazolo[3,4-*d*]pyrimidine (**15c**).

This compound (0.759 g, 2.9 mmoles) was synthesized in 98% yield from **14c** (0.954 g, 3 mmoles) and 30% hydrogen peroxide (10 ml) in a manner similar to that described for the preparation of **15a**. An analytical sample was recrystallized from methanol to give colorless needles, mp > 390°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3450, 3180, 3070 (OH or NH), 1751, 1650 (CO);  $^1\text{H}$  nmr (deuteriodimethyl sulfoxide):  $\delta$  3.82 (3H, s, OMe), 7.05 (2H, d,  $J = 9.0$  Hz, phenyl-H), 8.13 (2H, d,  $J = 9.0$  Hz, phenyl-H), 10.52 (1H, bs, NH or OH), 11.26 (1H, bs, NH or OH); ms:  $m/z$  258 ( $\text{M}^+$ , 9), 240 (12), 214 (13), 60 (31), 43 (100).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{10}\text{N}_4\text{O}_3$ : C, 55.81; H, 3.90; N, 21.70. Found: C, 55.44; H, 3.94; N, 21.37.

4-Benzylamino-6-methylthio-3-phenylpyrazolo[3,4-*d*]pyrimidine (**16a**).

A mixture of 0.864 g (3 mmoles) of **14a** and 0.643 g (6 mmoles) of benzylamine was heated at 150° for 2 hours. After cooling, the product was recrystallized from methanol to give 0.608 g (1.7 mmoles) of **16a** as colorless needles, mp 195°, in 58% yield; ir (potassium bromide):  $\nu$   $\text{cm}^{-1}$  3420 (NH); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 252 (4.52), 290 (4.13, shoulder).

*Anal.* Calcd. for  $\text{C}_{19}\text{H}_{17}\text{N}_5\text{S}$ : C, 65.68; H, 4.93; N, 20.16; S, 9.23. Found: C, 65.70; H, 5.07; N, 19.92; S, 8.90.

4-Benzylamino-3-(2-methoxyphenyl)-6-methylthiopyrazolo[3,4-*d*]pyrimidine (**16b**).

This compound (0.643 g, 1.7 mmoles) was synthesized in 66% yield from **14b** (0.954 g, 3 mmoles) and benzylamine (0.643 g, 6 mmoles) in a similar manner to that described for the preparation of **16a**. An analytical sample was recrystallized from methanol to give colorless needles, mp 193°, ir (potassium bromide):  $\nu$  max

$\text{cm}^{-1}$  3410 (NH), 1575; uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 249 (4.51), 286 (4.28);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.58 (3H, s, SMe), 3.49 (3H, s, OMe), 4.69 (2H, d,  $J = 6.0$  Hz,  $-\text{CH}_2-$ ), 5.71 (1H, bs, NH), 7.22 (5H, s, phenyl-H).

*Anal.* Calcd. for  $\text{C}_{20}\text{H}_{19}\text{N}_5\text{OS}$ : C, 63.64; H, 5.07; N, 18.55; S, 8.49. Found: C, 63.40; H, 5.02; N, 18.54; S, 8.70.

Methyl 2-Pyrrolyldithiocarboxylate (**17**).

A 500 ml round-bottomed three-necked flask was equipped with a mechanical stirrer with a suitable seal, an efficient reflux condenser protected from moisture in the air by a drying tube. The flask was surrounded by an ice bath. Approximately 20 ml of dry ether and 6.03 g (0.26 g-atom) of magnesium were placed in the flask and cooled to about 0°. Ethyl bromide (21.8 g, 0.20 mole) was slowly added to the above stirred mixture of ether and magnesium. The reaction mixture was then warmed until the reaction of ethyl bromide and magnesium was under way. A crystal of iodine may be added if the reaction does not start readily. The ethyl bromide was then allowed to distil into reaction mixture during a period of about 1 hour. The reaction mixture and the tube containing the ethyl bromide were cooled if the refluxing of the reaction mixture becomes so vigorous that the reflux condenser does not condense the ethyl bromide. After all the ethyl bromide has been added, the reaction mixture was warmed for one hour so that there was a gentle reflux. After cooling at 0°, a solution of 13.4 g (0.20 mole) of freshly distilled pyrrole in 20 ml of dry ether was added dropwise while the reaction mixture was stirred vigorously and cooled. The reaction mixture was stirred for 1 hour at room temperature. At this time, pyrrolylmagnesium bromide was prepared. Carbon disulfide (15.3 g, 0.20 mmole) was slowly added to the above Grignard reagent under stirring at 0° over a period of 20 minutes and stirring was continued for 1 hour at room temperature. To a Grignard reaction product (magnesium 2-pyrrolyldithiocarboxylate), 57.0 g (0.40 mole) of methyl iodide was slowly added dropwise while the reaction mixture was stirred vigorously at room temperature. After stirring for 3 hours, the reaction mixture was decomposed by adding 20 ml of a saturated ammonium chloride solution dropwise, with vigorous stirring, to the thoroughly cooled reaction. After the reaction mixture has been allowed to stand for one hour, the ether solution was poured off and the precipitate washed by decantation with two 200 ml portions of ether. After drying over sodium sulfate, the ether was removed by distillation, and the residual methyl 2-pyrrolyl-dithiocarboxylate was distilled. This product could be also purified by alumina column chromatography using hexane as an eluent to give a yellow oil. The yield of methyl 2-pyrrolyldithiocarboxylate, a yellow liquid boiling at 59°/13 mm Hg, was 14.7 g, (47%) [65];  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  (3H, s, SMe), 6.20-6.43 (1H, m, 3-H), 6.96-7.20 (2H, m, 4, 5-H), 8.90 (1H, m, NH).

2-Cyano-3-methylthio-3-pyrrol-2-ylacrylonitrile (**18**).

To a solution of 3.15 g (20 mmoles) of methyl 2-pyrrolyldithiocarboxylate (**17**) in 30 ml of benzene, 3.46 g (24 mmoles) of fresh tetracyanoethylene oxide (**9**) was slowly added in small portions under stirring at 0°. Stirring was continued for 2 hours at room temperature. After removal of the solvent under reduced pressure. The residue was chromatographed over by column using benzene as an eluent to give 3.33 g (17.6 mmoles) of yellow needles, mp 112° [lit [62] mp 112°], in 88% yield.

Dimethyl 4-Cyano-3-methylthio-5-aza[2.2.3]cycloazine-6,7-dicarboxylate (**20**).

A mixture of 0.946 g (5 mmoles) of **18** and 5 ml of triethylamine was refluxed for 1 hour. After removal of the excess of triethylamine, the product was heated at 150° for 1 hour to give the cyclized product. 2,6-Lutidine can also be used as catalyst of cyclization of **18** instead of triethylamine. This material was used in the next step without purification. A mixture of the above crude imino, 1.066 g (7.5 mmoles) of dimethyl acetylenedicarboxylate, 0.90 g of 5% of palladium-on-charcoal (Pd-C), and 30 ml of toluene was refluxed for 20 hours. After removal of the solvent and Pd-C, the residue was chromatographed on an alumina column using benzene as an eluent to give orange needles. This compound was recrystallized from methanol to give 0.378 g (1.15 mmoles) of **20** as orange needles, mp 212°, in 23% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2210 (CN), 1740, 1715 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 239 (4.45), 278 (4.28), 302 (4.26), 376 (4.32), 440 (3.61);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.03 (3H, s, SMe), 4.08 (3H, s, OMe), 4.10 (3H, s, OMe), 7.62 (1H, d, J = 4.5 Hz, 2-H), 7.86 (1H, d, J = 4.5 Hz, 1-H); ms: (m/z) 329 ( $\text{M}^+$ , 100), 298 (44), 270 (17), 240 (11), 213 (48).

*Anal.* Calcd. for  $\text{C}_{14}\text{H}_{11}\text{N}_3\text{O}_4\text{S}$ : C, 54.71; H, 3.37; N, 12.76; S, 9.74. Found: C, 54.52; H, 3.44; N, 12.70; S, 9.53.

#### 2-Cyano-3-morpholino-3-pyrrol-2-ylacrylonitrile (**21a**).

A solution of 0.189 g (1 mmole) of **18** and 0.174 g (2 mmoles) of morpholine in 10 ml of methanol was refluxed for 4 hours. After removal of the solvent, the residue was recrystallized from methanol to give 0.137 g (0.6 mmoles) of yellow needles, mp 164°, in 60% yield; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3280 (NH), 2200 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 244 (4.13, shoulder), 248 (4.14), 295 (4.30, shoulder), 303 (4.33), 314 (4.20, shoulder), 410 (3.64);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  3.94 (8H, s,  $\text{N-CH}_2\text{-CH}_2\text{-O}$ ), 6.27-6.40 (2H, m, 3,4-H), 7.40 (1H, dd, J = 1.1, 2.6 Hz, 5-H), 7.83 (1H, bs, 1-H); ms: (m/z) 228 ( $\text{M}^+$ , 100), 171 (16), 170 (14), 116 (17), 57 (12).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{12}\text{N}_4\text{O}$ : C, 63.15; H, 5.30; N, 24.55. Found: C, 62.99; H, 5.41; N, 24.27.

#### 2-Cyano-3-piperidino-3-pyrrol-2-ylacrylonitrile (**21b**).

This compound (0.145 g, 0.64 mmole) was synthesized in 64% yield from **18** (0.189 g, 1 mmole) and piperidine (0.170 g, 2 mmoles) in a manner similar to that described for the preparation of **21a**. An analytical sample was recrystallized from methanol to give yellow needles, mp 181°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3270 (NH), 2190 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 243 (4.14, shoulder), 249 (4.17), 292 (4.32, shoulder), 303 (4.36), 314 (4.22, shoulder), 4.04 (3.64);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  1.79 (6H, s, 3,4,5-H), 3.86 (4H, s, 2,6-H), 6.26-6.41 (2H, m, 3,4-H), 7.39 (1H, dd, J = 0.99, 2.75 Hz, 5-H), 7.78 (1H, bs, 1-H); ms: (m/z) 226 ( $\text{M}^+$ , 100), 197 (17), 84 (13).

*Anal.* Calcd. for  $\text{C}_{13}\text{H}_{14}\text{N}_4$ : C, 69.00; H, 6.24; N, 24.76. Found: C, 69.21; H, 6.52; N, 24.87.

#### 2-Cyano-3-pyrrolidino-3-pyrrol-2-ylacrylonitrile (**21c**).

This compound (0.161 g, 0.76 mmole) was synthesized in 76% yield from **18** (0.189 g, 1 mmole) and pyrrolidine (0.142 g, 2 mmoles) in a manner similar to that described for the preparation of **21a**. An analytical sample was recrystallized from methanol to give yellow needles, mp 190°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  3270 (NH), 2208 (CN); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 242 (4.18, shoulder), 249 (4.23), 290 (4.33, shoulder), 300 (4.37), 312 (4.20,

shoulder), 405 (3.65);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.03-2.25 (4H, m, 3,4-H), 3.70-3.98 (4H, m, 2,5-H), 6.25-6.32 (2H, m, 3,4-H), 7.39 (1H, s, 5-H), 7.73 (1H, s, 1-H); ms: (m/z) 212 ( $\text{M}^+$ , 100), 184 (27), 157 (11), 116 (12), 70 (16).

*Anal.* Calcd. for  $\text{C}_{12}\text{H}_{12}\text{N}_4$ : C, 67.91; H, 5.70; N, 26.40. Found: C, 67.87; H, 5.82; N, 26.08.

Dimethyl 4-Cyano-3-morpholino-5-aza[2.2.3]cycloazine-6,7-dicarboxylate (**23a**).

This compound (0.083 g, 0.23 mmole) was synthesized in 45% yield from **21a** (0.114 g, 0.5 mmole) and dimethyl acetylenedicarboxylate (0.107 g, 0.75 mmole) in a manner similar to that described for the preparation of **20**. An analytical sample was recrystallized from methanol to give yellow leaflets, mp 271°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2240 (CN), 1738, 1725 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 222 (4.46), 244 (4.17), 267 (4.26), 273 (4.26, shoulder), 297 (4.23), 324 (4.03, shoulder), 381 (4.33);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  4.03 (3H, s, OMe), 4.06 (3H, s, OMe), 3.90-4.04 (8H, m,  $\text{N-CH}_2\text{-CH}_2\text{-O}$ ), 7.33 (1H, d, J = 4.2 Hz, 2-H), 7.54 (1H, d, J = 4.2 Hz, 1-H); ms: (m/z) 368 ( $\text{M}^+$ , 100), 337 (21), 252 (26), 149 (21), 81 (31), 71 (21), 69 (57), 57 (50), 55 (27).

*Anal.* Calcd. for  $\text{C}_{18}\text{H}_{16}\text{N}_4\text{O}_5$ : C, 58.69; H, 4.38; N, 15.21. Found: C, 58.26; H, 4.46; N, 14.96.

Dimethyl 4-Cyano-3-piperidino-5-aza[2.2.3]cycloazine-6,7-dicarboxylate (**23b**).

This compound (0.132 g, 0.36 mmole) was synthesized in 72% yield from **21b** (0.113 g, 0.5 mmole) and dimethyl acetylenedicarboxylate (0.107 g, 0.75 mmole) in a similar manner to that described for the preparation of **20**. An analytical sample was recrystallized from methanol to give yellow needles, mp 253°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2205 (CN), 1742, 1721 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 223 (4.40), 242 (4.09, shoulder), 267 (4.25), 272 (4.24, shoulder), 297 (4.21), 377 (4.31);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  1.87 (6H, m, 3,4,5-H), 4.02 (4H, m,  $\text{N-CH}_2$ ), 4.05 (6H, s, OMe), 7.25 (1H, d, J = 4.2 Hz, 2-H), 7.50 (1H, d, J = 4.2 Hz, 1-H); ms: (m/z) 366 ( $\text{M}^+$ , 100), 309 (58), 250 (32), 81 (21), 69 (35).

*Anal.* Calcd. for  $\text{C}_{19}\text{H}_{18}\text{N}_4\text{O}_4$ : C, 62.29; H, 4.95; N, 15.29. Found: C, 61.93; H, 5.07; N, 14.99.

Dimethyl 4-Cyano-3-pyrrolidino-5-aza[2.2.3]cycloazine-6,7-dicarboxylate (**23c**).

This compound (0.078 g, 0.22 mmole) was synthesized in 44% yield from **21c** (0.106 g, 0.5 mmole) and dimethyl acetylenedicarboxylate (0.107 g, 0.75 mmole) in a manner similar to that described for the preparation of **20**. An analytical sample was recrystallized from methanol to give yellow needles, mp 236°; ir (potassium bromide):  $\nu$  max  $\text{cm}^{-1}$  2205 (CN), 1742, 1721 (CO); uv (ethanol):  $\lambda$  max nm (log  $\epsilon$ ) 222 (4.46), 244 (4.15), 265 (4.25), 271 (4.24), 297 (4.24), 325 (3.99), 378 (4.32);  $^1\text{H}$  nmr (deuteriochloroform):  $\delta$  2.18-2.33 (4H, m,  $\text{-CH}_2\text{-CH}_2\text{-}$ ), 4.02 (3H, s, OMe), 4.06 (3H, s, OMe), 3.80-4.20 (4H, m,  $\text{N-CH}_2$ ), 7.16 (1H, d, J = 4.2 Hz, 2-H), 7.41 (1H, d, J = 4.2 Hz, 1-H); ms: (m/z) 352 ( $\text{M}^+$ , 19), 236 (5), 144 (22), 69 (28), 57 (33), 44 (79), 43 (100).

*Anal.* Calcd. for  $\text{C}_{18}\text{H}_{16}\text{N}_4\text{O}_4$ : C, 61.36; H, 4.58; N, 15.90. Found: C, 61.10; H, 4.67; N, 15.77.

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## REFERENCES AND NOTES

- [1] For Part III: see, Y. Tominaga, Y. Matsuoka, and A. Hosomi, *Heterocycles*, **27**, 2791 (1988); for the previous papers in this series: Part I: Y. Tominaga, Y. Matsuoka, S. Kohra, and A. Hosomi, *Heterocycles*, **26**, 613 (1987); Part II: Y. Tominaga, Y. Ichihara, and A. Hosomi, *Heterocycles*, **27**, 2345 (1988).
- [2] A. Thuillier, *Phosphorus Sulfur*, **23**, 253 (1985).
- [3] F. Duus, "Thiocarbonyl Compounds" in "Organic Sulphur Compounds", "Comprehensive Organic Chemistry", 1979.
- [4] S. McKenzie, in "Organic Compounds of Sulphur, Selenium, and Tellurium", Vol I, D. H. Reid, ed, The Chemical Society, London, 1970, Chapter 5.
- [5] A. Ishida, H. Fujii, T. Nakamura, T. Oh-ishi, K. Aoe, Y. Nishibata, and A. Kinumaki, *Chem. Pharm. Bull.*, **34**, 1994 (1986).
- [6] G. Kobayashi, Y. Matsuda, Y. Tominaga, and K. Mizuyama, *Chem. Pharm. Bull.*, **23**, 2749 (1975).
- [7] M. Roth, P. Dubs, E. Gotshi, and A. Eschenmoser, *Helv. Chem. Acta*, **54**, 710 (1971).
- [8] T. Mukaiyama, T. Yamaguchi, and H. Nohira, *Bull. Chem. Soc. Japan*, **38**, 2107 (1965).
- [9] T. Yamaguchi, Y. Shimizu, and T. Suzuki, *Chem. Ind.*, **6**, 380 (1972).
- [10] R. Gompper and R. R. Schmidt, *Chem. Ber.*, **98**, 1385 (1965).
- [11] R. J. Sundberg, C. P. Walters, and J. D. Bloom, *J. Org. Chem.*, **46**, 3730 (1981).
- [12] S. Ruucher and P. Klein, *Tetrahedron Letters*, **21**, 4061 (1980).
- [13] Y. Tominaga, S. Kohra, and A. Hosomi, *Tetrahedron Letters*, **28**, 1529 (1987).
- [14] Y. Tominaga, Y. Matsuda, H. Hayashida, S. Kohra, and A. Hosomi, *Tetrahedron Letters*, **29**, 577 (1988).
- [15] J. Sandstrom, "Topics in Stereochemistry", Vol 14, N. L. Allinger, E. L. Eliel, and S. H. Wilen, eds, An Interscience Publication, John Wiley & Sons, New York, 1983, Chapter 2, pp 83-181.
- [16] Y. Tominaga and Y. Matsuda, *J. Heterocyclic Chem.*, **37**, 937 (1985).
- [17] Y. Tominaga and Y. Matsuda, *Yuki Gosei Kagaku Kyokai-Shi (J. Synth. Org. Chem., Japan)*, **43**, 669 (1985).
- [18] Y. Tominaga, Y. Shiroshita and A. Hosomi, *Heterocycles*, **27**, 2251 (1988).
- [19] Y. Tominaga, *Yuki Gosei Kagaku Kyokai-Shi (J. Synth. Org. Chem. Japan)*, **47**, 413 (1989).
- [20] R. K. Dieter, *Tetrahedron*, **42**, 3029 (1986).
- [21] D. L. Boger and M. D. Mullican, *J. Org. Chem.*, **49**, 4050 (1984).
- [22] A. Kakehi, S. Ito, K. Uchiyama, and K. Kondo, *J. Org. Chem.*, **43**, 2896 (1978).
- [23] L. Crombie and R. V. Dove, *J. Chem. Soc., Perkin Trans. I*, 686 (1979).
- [24] N. Takeuchi, K. Ochi, M. Murase and S. Tobinaga, *Chem. Pharm. Bull.*, **31**, 4360 (1983).
- [25] T. Severin, B. Bruck, and P. Adohicary, *Chem. Ber.*, **99**, 3097 (1966).
- [26] T. Severin and D. Konig, *Chem. Ber.*, **107**, 1499 (1974).
- [27] R. Gompper and R. R. Schmidt, *Chem. Ber.*, **98**, 1385 (1965).
- [28] T. Mukaiyama, T. Yamaguchi, and H. Nohira, *Bull. Chem. Soc. Japan*, **38**, 2107 (1965).
- [29] T. Yamaguchi, Y. Shimizu, and T. Suzuki, *Chem. Ind.*, **6**, 380 (1972).
- [30] A. Ishida, T. Nakamura, K. Irie, and T. Ohishi, *Chem. Pharm. Bull.*, **33**, 3237 (1985).
- [31] M. Santas, *Liebigs Ann. Chem.*, 179 (1988).
- [32] A. K. Gupta, H. Ila, and H. Junjappa, *Tetrahedron Letters*, **28**, 1459 (1987).
- [33] B. Myrboth, H. Ila, and H. Junjappa, *J. Org. Chem.*, **48**, 5327 (1983).
- [34] R. Gompper and W. Topfl, *Chem. Ber.*, **95**, 2861 (1962).
- [35] R. Gompper and H. Schaefer, *Chem. Ber.*, **100**, 591 (1967).
- [36] Methyl dithiocarboxylates **2** were synthesized by the treatment of the corresponding thiolonium salts with hydrogen sulfide. See, R. Mayer, S. Scheithauer, and D. Kung, *Chem. Ber.*, **99**, 1393 (1966).
- [37] W. J. Linn and R. E. Benson, *J. Am. Chem. Soc.*, **87**, 3657 (1965).
- [38] W. J. Linn, *J. Am. Chem. Soc.*, **87**, 3665 (1965).
- [39] A. Rouessage and J. Vialle, *Bull. Soc. Chim. France*, 2054 (1968).
- [40] W. J. Linn, O. W. Webster, and R. E. Benson, *J. Am. Chem. Soc.*, **87**, 3651 (1965).
- [41] W. J. Linn, *Org. Synth.*, **5**, 1007 (1973).
- [42] D. J. Brown, "Pyrimidines and Their Benzo Derivatives", in "Comprehensive Heterocyclic Chemistry", Vol 3, A. R. Katritzky and C. W. Rees, eds, Pergamon Press, Oxford, 1984, pp 57-156.
- [43] E. C. Taylor and A. McKillop, "The Chemistry of Cyclic Enaminonitriles and *o*-Aminonitriles", in "Advances in Organic Chemistry: Methods and Results", Vol 7, E. C. Taylor, ed, Interscience Publishers, New York, 1970.
- [44] S. Kobayashi, *Chem. Pharm. Bull.*, **21**, 941 (1973).
- [45] R. K. Robins, *J. Am. Chem. Soc.*, **78**, 784 (1956).
- [46] C. C. Cheng and R. K. Robins, *J. Am. Chem.*, **21**, 1240 (1956).
- [47] Y. Tominaga, S. Kohra, H. Honkawa, and A. Hosomi, *Heterocycles*, 1409 (1989).
- [48] A. Bendich, D. J. Russell, Jr., and J. J. Fox, *J. Am. Chem. Soc.*, **76**, 6073 (1954).
- [49] R. Gompper and W. Topfl, *Chem. Ber.*, **95**, 2881 (1962).
- [50] E. C. Taylor, A. McKillop, and R. N. Warrenner, *Tetrahedron*, **23**, 891 (1967).
- [51] R. L. Schriener and F. W. Newmann, *Chem. Rev.*, **35**, 351 (1944).
- [52] H. Yamanaka and H. Sakamoto, *Yuki Gosei Kagaku Kyokai-Shi (J. Synth. Org. Chem. Japan)*, **43**, 951 (1985).
- [53] J. Kobe, R. K. Robins and D. E. O'Brien, *J. Heterocyclic Chem.*, **11**, 199 (1974).
- [54] A. Taurins, *Chem. Heterocyclic Compounds*, **30**, 271 (1977).
- [55] K. Matsumoto, T. Uchida, and J. Yamauchi, *Yuki Gosei Kagaku Kyokai-shi (J. Synth. Org. Chem. Japan)*, **35**, 793 (1977).
- [56] W. Flitsch and U. Kramer, "Advances in Heterocyclic Chemistry", Vol 22, A. Katritzky and A. J. Boulton, eds, Academic Press, New York, 1978, p 321.
- [57] W. Flitsch, "Pyrroles with Fused Six-Membered Heterocyclic Rings", (i) a-Fused, in "Comprehensive Heterocyclic Chemistry", Vol 4, A. R. Katritzky and C. W. Rees, eds, Pergamon Press, Oxford, 1984, p 443.
- [58] M. A. Jessep and D. Leaver, *J. Chem. Soc., Perkin Trans. I*, 1319 (1980).
- [59] V. Boekelheide and S. S. Kertelj, *J. Org. Chem.*, **28**, 3212 (1963).
- [60] M. A. Jessup and D. Leaver, *J. Chem. Soc., Chem. Commun.*, 790 (1970).
- [61] Y. Tominaga, Y. Matsuoka, S. Kohra, and A. Hosomi, *Heterocycles*, **26**, 613 (1987).
- [62] Heating at high temperature may be essential for the cyclization of **18**. Hartke and Radau reported the preparation of **18** by the Grignard reaction of pyrrole and ketene dithioacetal, 2,2-dicyano-1,1-bis(methylthio)ethylene, in rather low yield and conversion of **18** to **19**. However, when the author attempted the reaction of **19**, thus obtained, with DMAD, only the conjugate addition product of the N-H bond of pyrrole moiety of **18** to DMAD, but not the corresponding cycloadduct, was obtained. See, K. Hartke and S. Radau, *Liebigs Ann. Chem.*, 2110 (1974).
- [63] Y. Tominaga, Y. Shiroshita, T. Kurokawa, Y. Matsuda, and A. Hosomi, *J. Heterocyclic Chem.*, **25**, 185 (1988).
- [64] W. J. Linn, *Org. Synth.*, Vol 5, John Wiley & Sons, New York, 1973, p 1007.
- [65a] J. Houben and H. Pohl, *Chem. Ber.*, **40**, 1725 (1925); [b] J. M. Beiner, A. Thuillier, and M. G. Champetier, *C. R. Acad. Sci. Paris*, **274**, 642 (1972).